



# Commiphora mukul Nanoparticulate Drug Delivery System for Rheumatoid Arthritis: Design, Development, Characterization, and Preclinical Investigation

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## Abstract:

The study's objective is to evaluate the improved therapeutic efficacy and decreased toxicity of Commiphora mukul nanoparticulate drug administration for rheumatoid arthritis. The active chemical components of Commiphora mukul were initially separated via extraction and column chromatography. Additionally, Commiphora mukul extract nanoparticles were prepared and assessed using various metrics and methods following a positive assessment of the nanoparticles, Tablet formulations were developed and tested for anti-rheumatoid arthritis activity, stability, *in vitro* drug release and different evaluation parameters. Particle size analysis, SEM, and TEM results indicated that the produced nanoparticles had the appropriate size and morphology. The acute toxicity study was performed, and it suggested that the prepared nanoparticles had less toxicity. According to the stability study, they had a 3-month shelf life and a 90% drug entrapment rate. After 60 minutes, all 12 batches of Tablet formulations released more than 70% of the medication they contained, while five batches released more than 90%. According to the accelerated stability study, the initial drug release from the best batch was 92.5%, and six months later, it was 91.1%. While other parameters like thickness, weight variation, hardness, and friability continue to fall within acceptable ranges with minimal modifications. The result of preclinical research revealed that prepared nanoparticulate formulations responded better than standard and control formulations. According to anti-rheumatoid arthritis activity tests,

the produced formulation had improved therapeutic efficacy and decreased toxicity. Patients with rheumatoid arthritis can benefit more from this formulation.

**Keywords:** Rheumatoid arthritis, Commiphora mukul, Nanoparticles, Drug delivery, Preclinical study, Drug development

## 1. Introduction

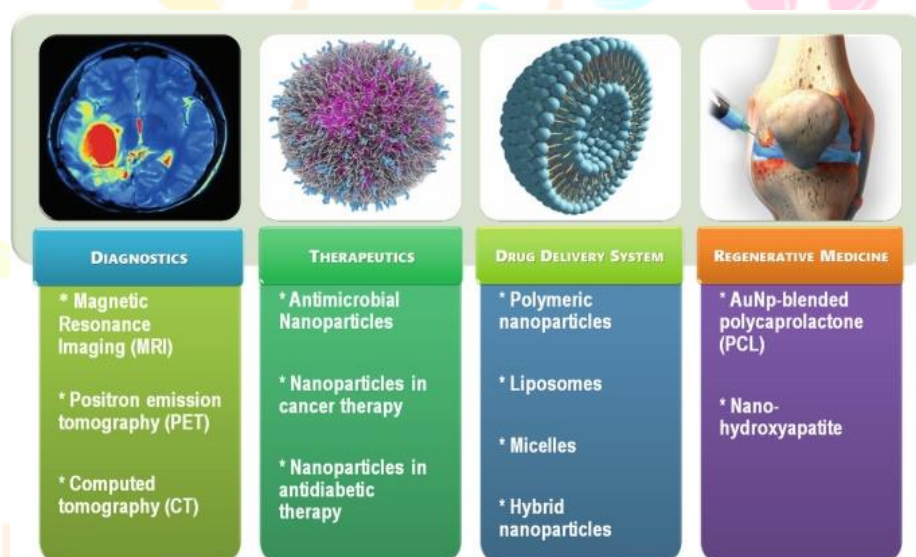
A series of chronic illnesses known as arthritis are characterized by joint inflammation and stiffness. It can result in discomfort, edema, limited movement, and a lower quality of life and affects millions of individuals worldwide. There are various kinds of arthritis, including psoriatic, rheumatoid, gouty, and osteoarthritis (the most prevalent variety) [1] [2] [3]. There are some medications available to treat this medical condition, however many of them have toxicity, drug delivery, and pharmacokinetic problems [4] [5]. In other hand a dose-dependent method of incorporating bioactive chemical components demonstrated remarkably high efficacy for the treatment of various illnesses [6]. It was shown that several phytoconstituents have the capacity to target a variety of inflammatory mediators, including those that are actively implicated in the pathogenesis of rheumatoid arthritis, such as nitric oxide (NO), cytokines, chemokines, adhesion molecules, NF-k, lipoxigenase (LOXs), and arachidonic acid (AA) [7]. Different bioactive components found in Commiphora mukul stem may be crucial for tracking rheumatoid arthritis.

Every disorder must be treated with an efficient drug delivery system and with an effective drug treatment plan. These kinds of difficulties are easily overcome by making herbal nanoparticulate medication delivery [8] [9]. Numerous benefits of using nanoparticulate drug delivery were demonstrated, including improved drug targeting, sustained drug release, drug protection and stability, greater drug solubility, decreased systemic toxicity, etc [10]. The study of phenomena and the manipulation of materials at the atomic, molecular, and macromolecular scales where their properties are very different from those of larger scales known as nanoscience [11] [12] [13].

The design, characterization, manufacture, and use of structures, devices, and systems using nanoscale shape and size-control constitutes nanotechnology. The physical, chemical, and biological characteristics of materials at the nanoscale are fundamentally and significantly different from those of discrete atoms and molecules or bulk matter [14] [15]. Research and development in nanotechnology is focused on comprehending these novel features and developing better materials, tools, and systems to take advantage of them. Because of its distinct size (1-100 nm) and high surface-to-volume ratios, nanotechnology has the potential to provide solutions to the existing

challenges in cancer therapy. Because of the makeup of their materials, nanotechnologies have characteristics such as self-assembly, stability, specificity, drug encapsulation, and biocompatibility [16] [17] [18].

The use of nanotechnology in cancer prevention, detection, diagnosis, imaging, and treatment has significant potential [19] [20]. Nanotechnology involves viewing, measuring, modeling, and manipulating materials at this scale and encompasses nanoscale science, engineering, and technology. Aerospace, agriculture, biotechnology, homeland security and national defense, energy, environmental improvement, information technology, medicine, and transportation are just a few of the industries and technology sectors that nanotechnology has the potential to transform and revolutionize [21] [22] [23]. It is now possible to spot applications that will have an influence on the world we live in thanks to advances in some of these fields of discovery (Fig. 1).



**Fig. 1.** Current advances in Nanotechnology

In this investigation, we used column chromatography, TLC detection, and standard extraction methods to identify and separate certain active chemical compounds found in *Commiphora mukul*. The produced extract-loaded chitosan nanoparticles were then used to make tablets. The manufactured tablets were assessed using several assessment criteria before being consumed by rats to determine their anti-arthritis activity.

## 2. Materials and methods

### 2.1. Collection, authentication and drying of plant material

The *Commiphora mukul* were taken from the district of Pune (Maharashtra). The plant was authenticated by D. L. Shirodkar, Botanist, Botanical Survey of India, Pune by comparing morphological features. The herbarium of the plant specimen was deposited at Botanical Survey of India, Pune; with the Voucher specimen number CDP-01 (Ref. No. BSI/WRC/IDEN.CER. /2021/H3 Dated 04/06/2021). The plant material

was dried in the shade at room temperature ground into grinder and powder material was passed into 120 mesh size.

## 2.2. Chemicals, reagents & solvents

Petroleum ether, Ethanol, Methanol, Dichloromethane, n-hexane, Ethyl acetate, Glacial acetic acid, N-Butanol, Chloroform, Acetone, Formic acid, Benzene, Dimethyl sulfoxide (DMSO), conc. Sulphuric acid, Hydrochloric acid, Benzene, pyridine, toluene, anisaldehyde, calcium chloride, copper sulphate, Ferric chloride, Follin's reagent, Iodine, Lead acetate, Magnesium chloride, Mercuric chloride, Ninhydrin, Nitric acid, Phloroglucinol, Potassium iodide, Potassium Dichromate, Potassium sodium Tartarate, Ruthenium red, Sodium acetate, Sodium iodide, Sodium hydroxide, Sodium nitroprusside, Hide powder, Folin Ciocalteu reagent, Sodium bicarbonate, Gallic acid and all the chemicals and reagents are analytical grade (Research lab Fine Chemicals Pvt. Ltd Mumbai, SD Fine Chem Mumbai, and Merck, India) were purchased from local suppliers.

## 2.3. Standardization of plant material

Pharmaceutical evaluation is crucial for standardizing plant medicines. The Indian Pharmacopoeia 2014 and other standard reference books were used to standardize the *Commiphora mukul* stem. The standardization of herbal products is done in accordance with WHO guidelines and involves measuring the importance of the ash, the extractive value, and the amount of drying that is lacking. It is highly helpful to understand the morphology, microscopy, and physical characteristics of herbal remedies. Experiments in pharmaceuticals provide important research insights on the purity, consistency, and identification of plant drugs [24] [25] [26] [27].

The following study was performed to standardize the *Commiphora mukul* stem: A) Pharmacognostic Study, B) Physical Evaluation. The physical evaluation contains Determination of foreign organic matter, moisture content, Ash value, Total ash, Water-soluble ash, and Acid-insoluble ash. Finally, the study contains evaluation of extractive values like water-soluble and Alcohol-soluble extractive value.

## 2.3. Extraction

The targeted guggule sterons steroidal components of *Commiphora wightii* were extracted utilizing the continuous Soxhlet extraction method with hydroalcoholic solvent. 200 grams of the crude drug were extracted with an adequate quantity of hexane using a soxhlet extractor at 60°-70°C. (Boiling point: 69°C). The air-dried marc of crude drug was repacked in a soxhlet apparatus extracted with ethyl acetate at 60°-80°C. (Boiling point: 77°C). After ethyl acetate extraction, the air-dried marc of crude drug was repacked in a soxhlet apparatus and extracted with ethanol at 40-50°C. (Boiling point: 78°C). After ethanol extraction, the air-dried marc of crude drug was macerated with distilled water for 48 hours with stirring. Then the resultant extract was filtered through a muslin cloth and the marc was separated. The filtrate was evaporated to dryness on hot plate at 45°C

to get an aqueous semisolid extract. In accordance with the methods described above, an extract was evaluated for preliminary phytochemical screening and thin-layer chromatography for the identification of separated components.

#### 2.4. Thin layer chromatography of extract

Following a preliminary phytochemical analysis of the extract, thin layer chromatography was used to further assess it. **Table 1** lists the various solvent systems that were employed for the identification of chemical constituents and their detection.

**Table 1** Chemical constituents, solvent system, and their detection by TLC.

Sr. no.	Chemical constituent	Mobile Phase	Detection
1.	Alkaloids	n- butanol: Ethyl acetate: Formic acid: Water (30:50:10:10) Toluene: Ethyl acetate: Formic acid (50:40:10)	UV -365nm
2.	Glycoside	Ethyl acetate:Methanol: Water (100:16.5:13.5)	UV -365nm
3.	Flavonoid	Toluene: Ethyl acetate: Glacial acetic acid: Water (100:11:11:26)	Anisaldehyde – Sulfuric acid. UV -365nm
4.	Steroids	Toluene: Ethyl acetate (9: 1) Ethyl acetate:Methanol: Water (70:20: 10)	Vanillin – Sulfuric acid. Anisaldehyde- Sulphuric acid reagent
5.	Terpenoids and Carotenoids	Cyclohexane: Ethyl acetate (75: 25)	UV- 268nm
		Petroleum ether: Benzene (9: 1)	UV- 254nm

#### 2.5. Column chromatography of *Commiphora wightii* stem extract

Additionally, column chromatography was used to separate the desired chemical components of the *Commiphora wightii* stem extract. The chromatographic column in question had a diameter of 3 cm and a height of 40 cm. The stationary phase utilized was Silica Gel 60 (Mesh 230-400), while the solvent for packing the

columns used was n-Hexane. Gradient solvent systems range from non-polar to highly polar solvents. Different ratios of n-hexane, ethyl acetate, methanol, and water were utilized. Each portion had a 20 ml volume.

## 2.6. Acute toxicity study

Guidelines for oral acute toxicity research are governed by the Organization for Economic Co-operation and Development (OECD). Nine adult albino rats were separated into three groups of three for the acute toxicity study. For the evening, all the animals were fasting. Separately diluted in 1% CMC, the separated extracts of two different plants were administered orally at doses of 300, 1000, and 2000 mg/kg body weight, respectively. The animals were watched for any symptoms of mortality for two hours, and then again for four hours. The animals were kept under close observation for a further 14 days after the initial 72 hours of monitoring for gross behavior, pupil size, general motor activity, convulsion, water intake, feces output, writhing, and any other hazardous indications.

## 2.7. Preparation of *Commiphora mukul* extract loaded chitosan nanoparticles

Sodium tripolyphosphate (TPP) was used as a cross linker with a small modification to create the chitosan nanoparticles in accordance with the ionic gelation procedure. Chitosan was fully dissolved in acetic acid prior to the addition of the *Commiphora mukul* extract (5%) and magnetic stirring of the mixture. TPP (0.5%) was then gradually administered via syringe at a consistent pace. In this procedure, glacial acetic acid (1.6%) was dissolved in distilled water, and chitosan (0.5–1) was added as a concentration. After 2 hours of additional stirring, it was centrifuged for 10 minutes at 10,000 rpm. The remainder was re-dissolved in phosphate buffer saline (PBS) after supernatant was discarded. They gathered the nanoparticle. Before being used, the prepared nanoparticle was lyophilized and kept at 40°C. Particle size, polydispersity index, UV-visible spectroscopy, X-ray powder diffraction (XRD), Zeta potential, % Yield of nanoparticles, Drug entrapment efficiency, In-vitro drug release, and stability were studied to characterize the produced nanoparticles.

## 2.8. Herbal tablet formulation

### 2.8.1. Pre-compression study

Pre-compression testing was done on the powder before it was compressed into tablet form. Pre-compression parameters attest to the final dosage form's high quality. For the analysis of the desired powder's quality, the ensuing parameters were put to the test.

#### 2.8.1.1. Angle of Repose

The angle of repose can be used to calculate the frictional forces present in loose powder or granules. This is the greatest angle that can be formed between a pile of powder or grains' surface and the horizontal plane.

$$\mathbf{\tan\theta = h/r}$$

$$\mathbf{\theta = \tan^{-1} (h/r) \text{ Formula 1}}$$

Where,  $\theta$  is the angle of repose,  $h$  is the height,  $r$  is the radius.

#### 2.8.1.2. Bulk density (BD)

The ratio of a powder's mass to its bulk volume is known as bulk density. The distribution of particle size, shape, and the propensity of the particles to stick to one another are the main determinants of a powder's bulk density.

$$\mathbf{\text{Bulk density} = \text{weight of powder} / \text{Bulk volume.}}$$

$$\mathbf{D_b = M/V_0}$$

$M$  = mass of the powder;  $V_0$  = bulk volume of the powder.

#### 2.8.1.3. Tapped density (TD)

It is the proportion of the powder's total mass to its tapped volume.

$$\mathbf{\text{Tapped density} = \text{Weigh of powder} / \text{Tapped volume}}$$

$$\mathbf{D_t = (M) / (V_t)}$$

$M$  = mass of the powder;  $V_t$  = tapped volume of the powder.

#### Carr's Index

Evaluation of a powder's BD, TD, and rate of packing down is a straightforward test. The following is the formula for Carr's index:

density Tapped - Bulk density

$$\text{Compressibility index} = 100 \times \frac{\text{density Tapped} - \text{Bulk density}}{\text{density Tapped}}$$

#### 2.8.1.4. Hausner's ratio

The Hausner's ratio is a proximate indicator of powder flow simplicity. The formula used to calculate it is as follows.

$$\text{Hausner's ratio} = \frac{\text{tapped density}}{\text{bulk density}}$$

#### 2.8.2. Tablet Evaluation

For the manufacture of compressible tablets, previously prepared *Commiphora mukul* nanoparticles were lyophilized. The 200 mg *Commiphora mukul* nanoparticles with varying composition ratios of HPMC K4M, HPMC K15M, MCC PH102, magnesium stearate, and Talc were used to make the 300 mg tablet utilizing the direct compression method. Total F1 to F12 (12 batches) were created utilizing various **Table 2** compositions.

**Table 2** Batch F1 to F12 with different composition of tablet.

Ingredients / Batch	<i>Commiphora mukul</i> (mg)	HPMC K4M (mg)	HPMC K15M (mg)	MCC PH102 (mg)	Magnesium stearate (mg)	Talc (mg)
<b>F1</b>	200	10	-	77	5	8
<b>F2</b>	200	20	-	67	5	8
<b>F3</b>	200	30	-	57	5	8
<b>F4</b>	200	-	10	77	5	8
<b>F5</b>	200	-	20	67	5	8
<b>F6</b>	200	-	30	57	5	8
<b>F7</b>	200	40	-	47	5	8
<b>F8</b>	200	50	-	37	5	8
<b>F9</b>	200	60	-	27	5	8



<b>F10</b>	200	-	40	47	5	8
<b>F11</b>	200	-	50	37	5	8
<b>F12</b>	200	-	60	27	5	8

#Total weight of compressed tablet- ±300mg

### 2.8.3. Post compression study

Compressible *Commiphora mukul* tablets (batches F1–F12) are analyzed for post–compression characteristics after production in order to assess tablet quality. The post-compression evaluation research was conducted using the aforementioned criteria.

#### 2.8.3.1. General appearance

The formed tablets general appearance was evaluated, and observations of shape, color, and texture were made.

#### 2.8.3.2. Weight Variation test

For more than 324 mg of tablets, no more than two of the individual weights departed from the average weight by more than 5.0%.

**Average weight** = weight of 20 tablets/20

$$\% \text{ Weight variation} = \frac{\text{Average waight} - \text{Weight of each tablet}}{\text{Average weight}} \times 100$$

Formula 3

#### 2.8.3.3. Thickness

Thickness of the tablets (n=3) was determined using a Vernier Caliper.

#### 2.8.3.4. Hardness test

Using the Monsanto hardness tester (n=3), the lower plunger was brought into contact with the tablet, and a reading of zero was taken to determine the tablets hardness. The tablet eventually broke when the plunger was turned by a threaded bolt against a spring. A pointer rode along a gauge in the barrel to show the force when the spring was squeezed.

#### 2.8.3.5. Friability test

This test is run to see whether tablets can tolerate abrasion during handling, packing, and transportation. It should ideally range from 0.5 to 1.0%.

$$\% \text{Friability} = [(W1 - W2) / W1] \times 100$$

Where, W1= weight of tablets before test, W2 = weight of tablets after test

#### 2.8.3.6. Assay of Tablet (Drug content)

First 20 tablets, weighted and finely powdered. The powder was then properly measured and transferred into a 100ml volumetric flask, equaling around 10mg of an herbal tablet. After that, a small amount of methanol was added, and the solution was sonicated for 30 minutes. Then, using methanol, the volume was adjusted to the mark. The clear solution (100ppm Stock solution) was obtained by filtering the solution via Whatmann filter paper. To make 8ppm, 10ppm, and 12ppm solutions, 0.8ml, 1.0ml, and 1.2ml of the stock solution (100ppm solution) were withheld. Utilizing methanol as a blank, the absorbance was determined spectrophotometrically by sweeping wavelengths between 400nm and 200nm. To help with the analysis, the absorbance was noted.

#### 2.8.3.7. In vitro Dissolution Study

According to USP, dissolving research was carried out on the prepared batches F1–F12. The USP dissolving Device II was used to determine the dissolving profile (F1-F12) in 900 ml of simulated fluid (7.4 pH Phosphate buffer) at a stirring speed of 50 rpm. At 0.5, 1, 2, 6, 8 and 12 hours, various aliquot samples were obtained with simulated substitute fluid in the same amount. A UV visible spectrophotometer was used to measure absorbance in order to determine how much medicine was discharged.

#### 2.8.4. In-vivo Anti-arthritis Study

In this investigation, Wistar albino rats were utilized, and they were purchased from Crystal Biological Solution in Pune, India. When the rats were treated at the age of seven weeks, their weight variance did not surpass 20% of the mean body weight for each sex. Rats were given the prescribed dosage orally. The Akindele and Adeyemi approach was initially used to create the arthritis syndrome, but with a few minor alterations. On the first and third days of the experiment, Wistar albino rats were subcutaneously injected with 0.1 ml (2.5% v/v in normal

saline) formaldehyde solution in the subplantar region of the right hind paw.

The rats were divided into five groups(n=5in each group) as follows:

Group I: Normal

GroupII:controlwhichreceivedaqueoussolution.

GroupIII:received10mg/kgDiclofenacsodium(PO)

GroupIV:receivedCMN(PO)

GroupV:receivedWSN(PO)

All groups received oral medication every day for 10 days, starting an hour before formaldehyde injection.

Rats' paw thicknesses were measured using a digital caliper on days 0, 2, 4, 6, and 8. Last but not least, radiograph recording and estimate of hemoglobin, C-reactive protein, and rheumatoid factor were carried out.

### 3.RESULT AND DISCUSSION

#### 3.1 Standardization of Plant Material

##### 3.1.1. Pharmacognostic Study and Physical characteristics

Selected *Commiphora mukul* plant parts underwent examination for their organoleptic characteristics, additional characteristics, and macroscopical details. (C. K. Kokate's Practical Book) *Commiphora mukul* is shrub or small tree, upto a maximum height of 4 m to 4.5m (13-15 ft).. The branches are thorny. The leaves are simple or trifoliate, the leaflets ovate, 1–5 cm (0.35–1.80 in) long, 0.5–2.5 cm (0.20–0.95 in) broad, and irregularly toothed. It secret brown coloured sticky exudate. Collection of Stem part with exudate used for further analysis (**Fig. 2**).



**Fig. 2. *Commiphora mukul* plant parts**

### 3.1.2. Physicochemical Evaluation

The foreign organic matter, moisture content, total ash value, water-soluble ash value, acid-insoluble ash value, water-soluble extractive value, and alcohol-soluble extractive value of *Commiphora mukul* stem were all evaluated. The results are shown as mean SEM in **Table 2** and are based on the observations. The amount of foreign stuff was found to be extremely minimal in the current study. For the purpose of obtaining desired quality phytochemicals, the moisture content present in crude medicine must be consistent and within the established range because it might alter microbial growth, enzyme activity, and the quality of plant material. The plant material's moisture content was fixed after the removal of the watery quantity shown in **Table 3**. The selected plant portions' moisture content was below the permissible level of 5%, which would prevent the growth of bacteria and fungi.

When assessing the quality and purity of pharmaceuticals, such as the presence or absence of foreign inorganic materials like silica, ash value is very crucial. The weight of ash in some medications varies very little, in terms of percentage, from sample to sample; any significant variation denotes a change in quality. The metrics that can be used to validate and standardize the characteristics of *Commiphora mukul* stem are their water solubility, acid-insolubility, and sulphated ash value. The chemical components of powdered medications are evaluated using their extractive values, which also aids in determining which chemical components are soluble in each

solvent. More components are found to be soluble in alcohol so the alcohol soluble extractive value is greater than the water-soluble extractive value.

**Table 3** Physicochemical analysis observations.

Evaluation parameters	Observations
Foreign organic matter	0.8± 0.20
Moisture content	3.09 ±0.04
<b>Ash values</b>	
Total ash value	09.0± 0.41
Water-soluble ash value	2.1±0.27
Acid-insoluble ash value	2.11±0.43
<b>Extractive values</b>	
Water soluble extractive value	6.86±0.42
Alcohol soluble extractive value	26.10±0.17
<b>Values are mean ± SEM, (n=3)</b>	

### 3.2. Extraction

*Commiphora mukul* stem were extracted using a hexane,ethyl acetate,ethanol& water solvent, and the ethanol extracts contained 21.01% & ethyl acetate extract contained 18.20%. of the required components by weight (**Table 4**). Therefore, we can draw the conclusion that the majority of phytochemicals may exhibit improved solubility in pertinently chosen solvents. Purification of isolated chemicals necessitates further separation.

**Table 4** Characteristics of *Commiphora wightii* stem extract.

Sr. No.	Extraction solvent used	Label	Percent Yield (% W/W)	Colour
1.	Hexane extract	CMHX	06.21%	Dark brown
2.	Ethyl acetate extract	CMEA	18.20%	Brown
3.	Ethanol extract	CMET	21.01%	Black
4.	Aqueous (Water) extract	CMAQ	10.15%	Green

#### 3.2.1. Preliminary Phytochemical Screening of Extract

Qualitative analysis of *Commiphora wightii* stem extract was performed to identify different phytoconstituents by using different qualitative tests and the results are depicted in **Table 5**.

**Table 5** Different qualitative tests performed to identify phytoconstituents in extract.

Sr.No.	Tests	<i>Commiphora wightii</i>			
		CMHX	CMEA	CMET	CMAQ
<b>Test for carbohydrate</b>					
1	Molish's test	-	-	-	+
	Benidicts test	-	+	+	+
	Fehling test	-	-	+	-
	Barfoed test	-	+	-	+
<b>Test for Proteins</b>					
2	Biuret Test	-	-	-	+
	Millions Test	-	-	-	-
<b>Test for amino acids</b>					
3	Ninhydrine test	-	+	-	+
<b>Test for Steroids</b>					
4	Salkowski test	-	+	-	-
	Libermann test	-	-	+	-
	Libermann-Burchard reaction	+	+	-	-
<b>Test for Glycosides</b>					
5	Anthraquinone glycoside test	-	+	+	-
	Cardiac glycoside: test	-	+	-	-
6	<b>Test for Terpenoids:</b>	-	+	-	+
<b>Test for Saponin</b>					
7	Foam test	-	+	-	-
<b>Test for Alkaloids</b>					
8	Dragondorff's test	-	+	-	-
	Mayer's test	-	+	+	-
	Hager's test	-	-	-	-
	Wagner's test	-	-	-	-
<b>Test for Tannins and Phenolic compounds</b>					
9	5% FeCl <sub>3</sub> test	-	+	-	+
	Lead acetate solution	+	-	+	-
<b>Test for Flavonoids</b>					

10	Shinoda test	-	+	+	
	Lead acetate test	-	+		+
	Sodium hydroxide test	-	-	+	-

*Commiphora wightii* stem hexane extracts revealed the presence of steroids, tannins, ethyl acetate extract showed presence of glycosides, alkaloids, phenolic compounds, steroids as well as flavonoids, ethanol extract showed presence of glycosides, alkaloids, Phenolic compounds as well as flavonoids, aqueous extracts showed presence of amino acids, proteins, phenolic compounds and saponins. To further identify and separate out the steroidal components, from ethyl acetate and ethanol extract of *Commiphora wightii* stem a TLC and column chromatography analysis is required.

### 3.2.2. Thin layer chromatography

After extracts underwent a preliminary phytochemical examination, thin layer chromatography was used to record additional findings. The various extract elements were identified using various mobile phase percentages. **Table 6** provided the details.

**Table 6** TLC-Characterization, used mobile phase and observations.

Sr. no.	Chemical constituent	Mobile Phase	Observation	
			Extract-Rf values	
1.	Alkaloids	n- butanol : Ethyl acetate: Formic acid : Water (30:50:10:10) Toluene: Ethyl acetate: Formic acid (50:40:10)	CMHX	-
			CMEA	0.45
			CMET	0.52
			CMAQ	-
2.	Glycoside	Ethyl acetate : Methanol : Water ( 100 : 16.5 : 13.5 )	CMHX	-
			CMEA	0.60
			CMET	0.62
			CMAQ	-
3.	Flavonoid	Toluene : Ethyl acetate : Glacial acetic acid :Water (100:11:11:26)	CMHX	-
			CMEA	0.45
			CMET	0.70
			CMAQ	
4.	Steroids	Toluene: Ethyl acetate (9: 1)	CMHX	-
			CMEA	0.52

		Ethyl acetate : Methanol : Water (70 : 20 : 10)	CMET	0.55
			CMAQ	-
5.	Terpenoids and Carotenoids	Cyclohexane : Ethyl acetate ( 75 : 25)	CMHX	-
			CMEA	0.51
			CMET	-
			CMAQ	-
		Petroleum ether : Benzene ( 9 : 1)	CMHX	-
			CMEA	-
			CMET	-
		CMAQ	-	

### 3.2.3 Fractionation of *Commiphora mukul stem ethyl acetate* extract by column chromatography

After column chromatography was used to separate the various chemical components of *Commiphora mukul* stem that had been identified by TLC. Most polar *ethyl acetate* solvent was used to start the constituent separation process.. Ethyl acetate and benzene were used for the separation between them. Five fractions in all, each with a capacity of 100 ml, were collected. All fractions were concentrated, and their steroid and alkaloid content were assessed using thin layer chromatography. **Table 7** lists the various mobile phases, fraction codes, colors, and yield percentages for each fraction.

**Table 7** Separation of chemical constituents of *Commiphora mukul* stem extract with different solvent system.

Solvent used	Ratio	Fraction Code	Colour	% Yield
Ethyl acetate	100%	A1	yellow gum	0.4
Ethyl acetate: Benzene	7:3	A2	Dark brown	1.8
Ethyl acetate: Benzene	5:5	A3	Dark Brown	1.6
Ethyl acetate: Benzene	2:8	A4	Brown	1.6
Benzene	100%	A5	Brown	2

### 3.3.Characterizations of Nanoparticles

The prepared nanoparticles of *Commiphora mukul* (*fraction 2*) were subjected for different evaluations parameters.

#### 3.3.1. Particle size determination by Zeta sizer

The size of the nanoparticles created from *Commiphora mukul* extract was measured. The size of the nanoparticles' particles was estimated to be between 1 and 100 nm. The tested nanoparticles' average particle size was discovered to be 43.1 nm. These nanoparticles' zeta potential was discovered to be -28.45 mV (**Table**

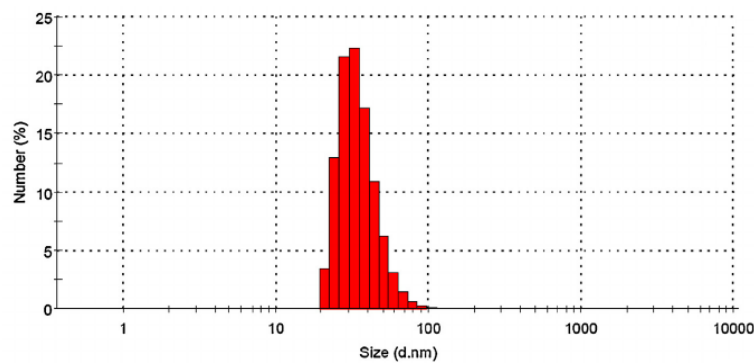


8). The produced nanoparticles appear to be stable, according to this. **Fig. 3** and **4** displayed the peak of the zeta particle size distribution and the particle density index.

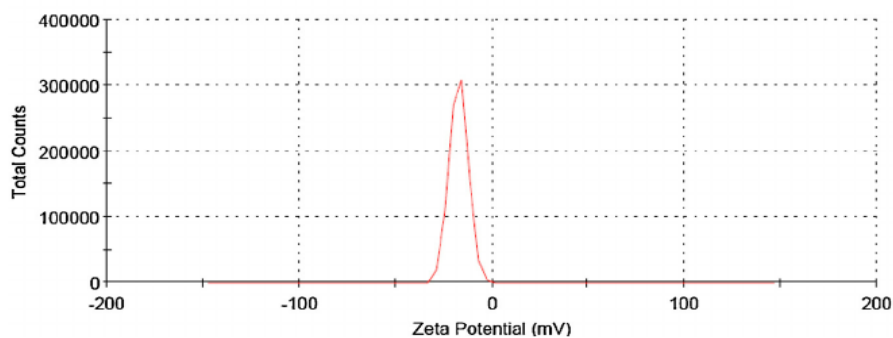
**Table 8** Particle size and zeta potential of extract of *Commiphora mukul*.

Sr. No.	Sample	Nanoparticle Size (nm)	Zeta Potential (mV)
1	Extract of <i>Commiphora mukul</i>	43.1 ± 12	-28.45

Values are shown as the mean ± standard deviation; n=5.



**Fig. 3.** Results of Particle Density Index of extracts derived extract of *Commiphora mukul* loaded nanoparticles.



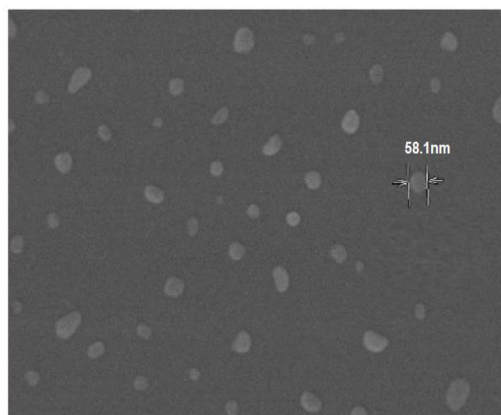
**Fig. 4.** Zeta particle size distribution peak of nanoparticles of extract of *Commiphora mukul*

### 3.3.2. Scanning electron microscopy

In order to analyze the surface morphology of silver nanoparticles, scanning electron microscopy was employed. The investigation helped the researchers better understand the morphological characteristics of the nanoparticle. There were several roughly spherical nanoparticles present, and they will tear apart from one another. A tiny, spherical nanoparticle with a small size can be seen in the SEM image of a freeze-dried silver nanoparticle with a longer cross-linking duration. Almost all the nanoparticles were sphere-shaped. *Commiphora mukul* extract produced nanoparticles with average diameters of 58.1nm nm. After being freeze-dried, the

nanoparticle dispersion formed sponge-like structures. The morphology of the sponge was ascertained by SEM

(Fig. 5).



**Fig. 5.** Scanning electron micrograph of nanoparticles obtained by extract of *Commiphora mukul*

### 3.3.3. Production yield of nanoparticles

The Production yield of prepared nanoparticles was calculated, and it was found 72.48%.

**Table 9** Production yield of all nanoparticles

Formulation	Production yield (%)
Extract of <i>Commiphora mukul</i>	72.48

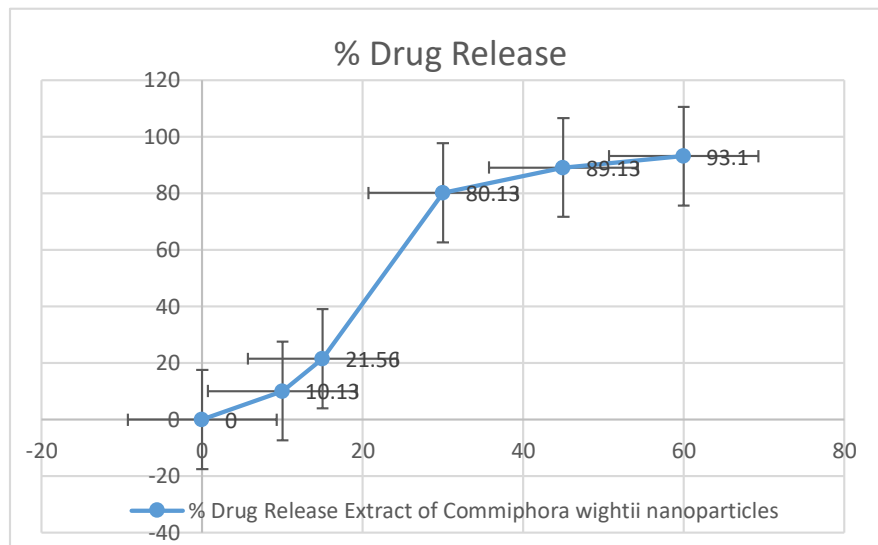
### 3.3.4. In-vitro release study

Drug release from *Commiphora mukul* extract-loaded nanoparticles was investigated in vitro. The maximum drug release was determined to be between 90 and 95 percent for all nanoparticles. It was investigated how generated nanoparticles released in vitro at 37 °C in phosphate buffered saline (PBS) (PH 7.4). By measuring absorbance at 440 nm with a UV-visible spectrophotometer, the amount of medicine released was calculated. From the initial 0 to 60 min, the drug release was examined at various time intervals (**Table 10**). At the 60th minute, a drug release of 93.10 was discovered. It implies that the formulation's oral in vitro drug release profile is favourable. **Fig. 6** displayed the cumulative medication release as a percentage.

**Table 10** In-vitro % drug release study.

Time (In Min)	% Drug Release
0	0
10	10.13± 0.15
15	21.56±0.24

30	80.13±0.11
45	89.13±0.25
60	93.10±0.35



**Fig. 6.** %Cumulative drug release

### 3.3.5. Drug entrapment efficiency

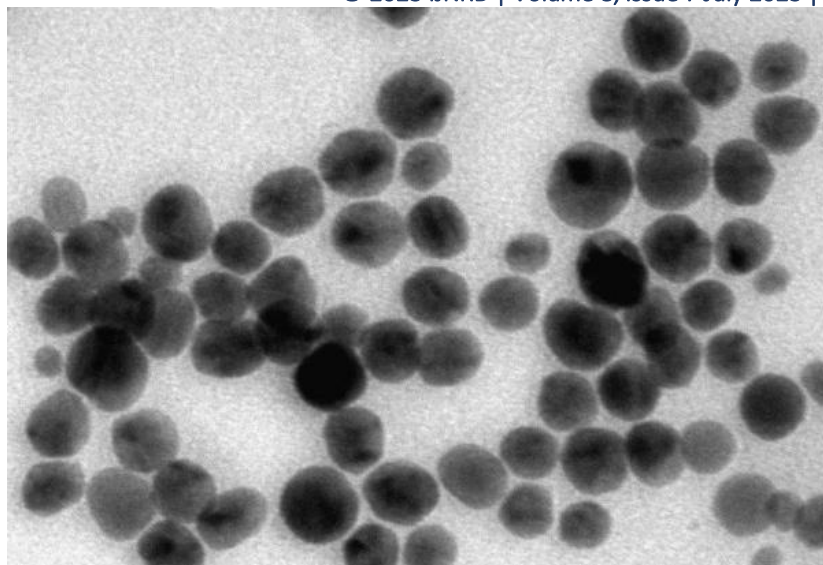
For formulation, the Commiphora mukul extract-loaded nanoparticles entrapment efficiency was found to be between 80% and 90%, showing higher drug entrapment efficiency (**Table 11**).

**Table 11** % Entrapment efficiency of formulation.

Formulations	Entrapment efficiency (%)
Extract of Commiphora mukul	86.12

### 3.3.6. Transmission electron microscopy

**Fig. 7** displayed a TEM micrograph of nanoparticles. According to the results of the TEM study, the average particle size was 45 nanometers, falling between the ranges of 30 and 60 nanometers (**Fig. 7**). The particles are round in shape. To confirm that no further types of metal oxide were present, the same sample was also examined using electron diffraction. Overall, the findings imply that the nanoparticles possess desirable qualities.



**Fig. 7.** TEM of Commiphora mukul nanoparticles

### 3.3.7. Stability of Nanoparticles

By examining the silver nanoparticles' absorption spectra after 12 weeks, the stability of the particles was assessed. The nanoparticles did not agglomerate and saw no significant changes during storage, suggesting that they were more stable. For prepared nanoparticles, the pattern of change in entrapment effectiveness, particle size, and zeta potential was the same. After three months of storage at 4°C, there was a modest (1%) increase in the size of the nanoparticles. Zeta potential was found to have decreased by 4%, whereas the entrapment efficiency of nanoparticles decreased by roughly 1% to 2%. The storage-related modifications that were noticed are insignificant (**Table 12**). The overall findings imply that the required nanoparticles stability was adequate.

**Table 12** Effect of storage on particle size, zeta potential and entrapment efficiency of Nanoparticles (n=3).

Values are expressed as mean  $\pm$ SD.

Storage time		“0” Month	“1” Month	“2” Month	“3” Month
<b>Particle size</b> (nm)	Extract F1	44.2 $\pm$ 10	44.5 $\pm$ 11	44.2 $\pm$ 09	43.1 $\pm$ 11
	Extract F3	44.3 $\pm$ 12	45.4 $\pm$ 12	45.6 $\pm$ 12	46.2 $\pm$ 12
<b>Zeta potential</b> (mV)	Extract F1	-29.32 $\pm$ 0.8	-28.58 $\pm$ 0.4	-27.88 $\pm$ 0.4	-26.17 $\pm$ 0.5
	Extract F3	-50.25 $\pm$ 0.5	-48.75 $\pm$ 0.8	-45.69 $\pm$ 0.5	-43.26 $\pm$ 0.7
<b>Entrapment efficiency (%)</b>	Extract F1	70.45 $\pm$ 0.3	71.24 $\pm$ 0.7	70.21 $\pm$ 0.7	69.54 $\pm$ 0.08
	Extract F3	74.25 $\pm$ 0.8	73.21 $\pm$ 0.3	72.45 $\pm$ 0.8	74.25 $\pm$ 0.5

(n=3). Values are expressed as mean  $\pm$ SD

The prepared lyophilized nanoparticle powder blends and herbal formulation (F1-F12) taken to formulate the Tablet and to carry out the pre formulation study.

### 3.4. Pre-formulation study

#### 3.4.1. Organoleptic studies

Powder of blend was found to be off-white.

#### 3.4.2. Precompression parameters

The details of different Pre compression evaluation parameters studied were reported in **Table 13**.

**Table 13** Different Pre compression evaluation parameters.

Batch	Angle of Repose ( $\theta$ )	Bulk Density (BD)	Tapped Density (TD)	Carr's Index	Hausner's Ratio
<b>F1</b>	28.21	0.41	0.42	2.38	1.02
<b>F2</b>	31.10	0.40	0.42	4.76	1.05
<b>F3</b>	27.01	0.41	0.42	2.38	1.02
<b>F4</b>	28.80	0.41	0.42	2.38	1.02
<b>F5</b>	30.24	0.41	0.43	4.76	1.05
<b>F6</b>	28.21	0.41	0.42	2.38	1.02
<b>F7</b>	30.81	0.41	0.42	2.38	1.02
<b>F8</b>	26.84	0.41	0.42	2.38	1.05
<b>F9</b>	26.56	0.4	0.42	2.38	1.05
<b>F10</b>	27.21	0.41	0.42	2.38	1.05
<b>F11</b>	27.32	0.41	0.42	2.38	1.02
<b>F12</b>	26.44	0.41	0.43	2.38	1.05



## 1.5 Evaluation of Tablets (Post compression parameters)

## 3.5.1 Organoleptic properties

All batches (F1-F12) were assessed for organoleptic properties like color, odor, and taste and found to be acceptable in all aspects.

## 3.5.2 General appearance

The formulated tablets were assessed for its general appearance and observations were made for shape, colour and texture (**Shape-** Round, **Colour-** off white, **Texture-** smooth). From the results obtained it was found that F1-F12 formulations have hardness, weight variation & friability within IP limit (**Table**

**14**).

**Table 14** Different Pre compression evaluation parameters.

<b>Batch</b>	<b>Weight Variation</b>	<b>Thickness Test</b>	<b>Hardness Test</b>	<b>Friability Test</b>	<b>Disintegration Test</b>
<b>F1</b>	3.12	4.49	5.5	0.64	12.15
<b>F2</b>	3.56	4.50	5.3	0.99	13.45
<b>F3</b>	3.15	4.50	5.5	0.66	12.12
<b>F4</b>	3.56	4.50	5.3	0.98	13.41
<b>F5</b>	3.25	4.49	5.4	0.66	12.20
<b>F6</b>	3.30	4.49	5.4	0.66	11.50
<b>F7</b>	3.41	4.50	5.4	0.66	11.12
<b>F8</b>	3.85	4.49	5.3	0.99	12.03
<b>F9</b>	3.45	4.50	5.5	0.66	11.56
<b>F10</b>	3.15	4.50	5.4	0.33	11.45
<b>F11</b>	3.69	4.50	5.3	0.33	11.42
<b>F12</b>	4.01	4.50	5.5	0.66	11.05

## 3.5.3 Dissolution Study

The corresponding Tables below contain information on the matrix tablets rates of dissolution. Each formulation's dissolution research was conducted in triplicate in 7.4 pH Phosphate Buffer.

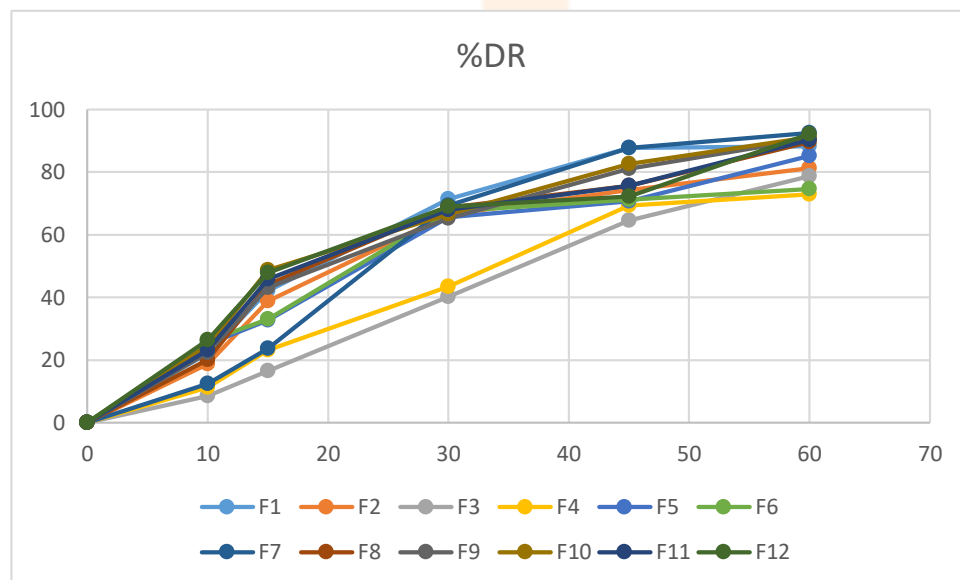
**Table 15** Cumulative drug release (F1-F12).

Time	F1	F2	F3	F4	F5	F6	F7
0	0	0	0	0	0	0	0
10	22.12±0.12	18.66±0.09	8.45±0.15	11.25±0.19	25.05±0.25	26.1±0.16	12.39±0.27
15	42.15±0.18	38.82±0.16	16.57±0.18	23.12±0.14	32.69±0.21	32.99±0.24	23.6±0.15
30	71.45±0.11	66.47±0.22	40.21±0.15	43.51±0.15	65.57±0.11	67.51±0.19	69.25±0.26
45	87.68±0.21	74.21±0.21	64.54±0.26	69.35±0.10	70.62±0.09	71.26±0.17	87.68±0.02
60	88.31±0.10	81.33±0.02	78.68±0.23	72.84±0.11	85.11±0.10	74.56±0.18	92.5±0.16

± S.D. n=6

Time	F8	F9	F10	F11	F12
0	0	0	0	0	0
10	20.12±0.36	22.36±0.63	24.56±0.48	23.14±0.63	26.32±1.63
15	44.15±0.48	43.12±1.05	48.74±1.52	45.78±1.59	47.88±0.85
30	68.59±0.89	65.25±1.15	66.53±1.05	67.89±1.56	68.95±0.48
45	75.62±1.05	81.15±0.59	82.56±1.04	75.66±0.54	72.15±0.47
60	89.56±1.03	90.56±0.14	91.14±0.63	90.21±1.04	92.74±0.52

± S.D. n=6



**Fig. 8.** % Cumulative drug release (F1-F12)

### 3.5.4 Stability Study (Accelerated study)

An accelerated stability study was carried out for Formulation F12 in compliance with ICH stability requirements. Hardness, overall drug release percentage, and weight variance. Parameters such as

friability were examined. The appearance, feel, and color of the produced tablets from batch F12 up until the stability period remained consistent. During the stability analysis, other parameters were found to be good. Investigation results revealed that the optimized formulation (F12) was stable.

**Table 16** Dissolution Study Data of stability study.

Time in Min	Initial	First month	Two month	Third month	Sixth month
0	0	0	0	0	0
10	12.39±0.27	12.10±0.24	12.16±0.95	13.19±0.14	12.24±0.65
15	23.6±0.15	23.63±0.59	22.86±0.49	23.01±0.19	23.46±0.79
30	69.25±0.26	70.11±0.14	69.16±0.74	70.66±0.64	69.29±0.64
45	87.68±0.02	86.36±0.36	86.48±0.76	86.76±0.16	87.32±0.66
60	92.5±0.16	92.9±0.14	91.01±0.58	91.17±0.23	91.10±0.14

**Table 17** Other parameters data of stability study.

Physical Parameter	Accelerated Stability Testing				
	Initial	First month	Two month	Third month	Sixth month
Appearance	Off white	Off white	Off white	Off white	Off white
Wt variation (%)	4.71	4.50	4.25	4.35	4.40
Hardness	5.4±0.13	5.4±0.23	5.3±0.16	5.4±0.21	5.4±0.18
Thickness	4.50±0.03	4.35±0.03	4.38±0.03	4.50±0.03	4.20±0.05
%friability	0.10±0.01	0.11±0.01	0.10±0.01	0.12±0.01	0.10±0.02

### 3.6 Anti-arthritis activity

Commiphora mukul nanoparticles were administered to normal, control, standard, and treated animals to determine the anti-arthritis activity in terms of paw volume. **Table 18** summarized the study's specifics. On days 0 through 10, the paw volume was documented in various groups, and responses were gauged. The percentage of paw volume inhibition was also measured on the tenth day, and it was discovered that the produced nanoparticle demonstrated 14.56% inhibition while the standard medication demonstrated 21%. According to the findings, the produced nanoparticles are more effective than conventional drugs.



**Table 18** Paw Volume on 0<sup>th</sup> to 10<sup>th</sup> day.

Groups	Treatment and Dose	Paw Volume (mm)					% Inhabitation of paw volume on 10 <sup>th</sup> Day
		0Day	2 <sup>nd</sup> day	4 <sup>th</sup> Day	6 <sup>th</sup> Day	8 <sup>th</sup> Day	
1	Normal	2.97±0.05	2.97±0.05	2.97±0.05	2.97±0.05	2.97±0.05	0
2	Control	2.65±0.08	5.92±0.17	8.621±0.34	8.54±0.19	7.57±0.34	-
3	Standard (Diclofenac)	2.79±0.18	5.31±0.26	7.72±0.34	5.5±0.35	4.91±0.42	21.03%
4	Test I CMN	2.69±0.24	6.84±0.54	7.33 ± 0.23	6.42 ± 0.24	5.19±0.47	14.56%

**On 4<sup>th</sup> Day**

**On 10<sup>th</sup> Day**



**Control**



**Standard**



## CMN

**Fig. 9.** Response of Control, standard and desired Commiphora mukul nanoparticles against animal groups

### 3.6.1 Hemoglobin Estimation

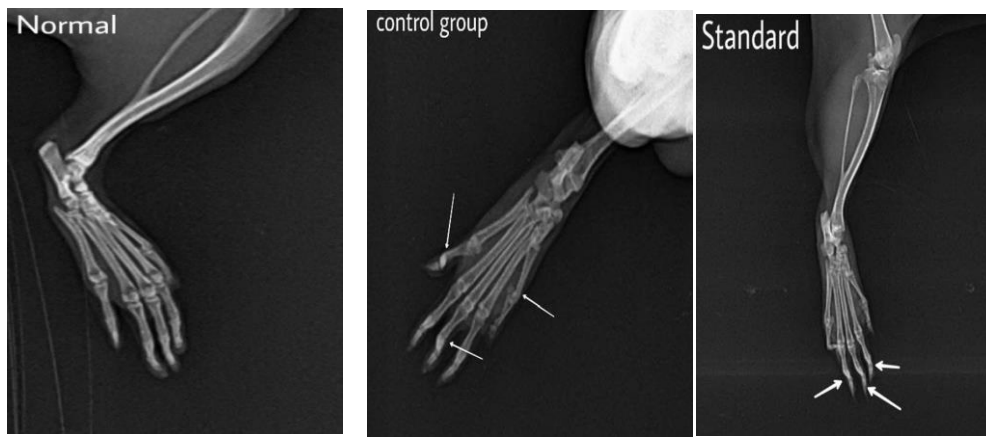
The findings of the hemoglobin estimation suggested that the usual medication and our nanoparticles produced quite comparable types of results. Both instances demonstrated that the animals' elevated hemoglobin levels were higher than those of the normal and control groups. **Table 19** contains information about it in depth.

**Table 19** Hemoglobin Estimation in different groups of animals.

Groups	Treatment andDose	Hbmg/dl
1	Normal	14.83±0.568
2	Control	10.5±0.856
3	Standard (Diclofenac)	16.03±0.254
4	CMN	14.23±0.556

### 3.6.2 Radiograph

On the basis of radiographs and coned-down views of the lower limbs, radiographic examination was carried out. In order to take radiographs, GE 500 mA, 40 kvp, and 4 MAS were used.



Radiological analysis revealed that the conventional and CMN treated animals had improved conditions.

**Fig. 10.** Radiograph of animals after and before the treatment

### 3.6.3 Estimation of C - reactive protein

A diagnostic kit is used to estimate the biochemical parameter. C-reactive protein (CRP) test kit estimation. To measure the degree of inflammation, C-reactive protein (CRP) in serum was detected in vitro. The C-reactive protein (CRP) levels in the CMN-treated animals were lower than those in the normal, control, and standard. **Fig. 11** depicted the effects of C-reactive protein (CRP) in more depth. **Table 20** displayed the effects of dosage therapy on several animal categories.



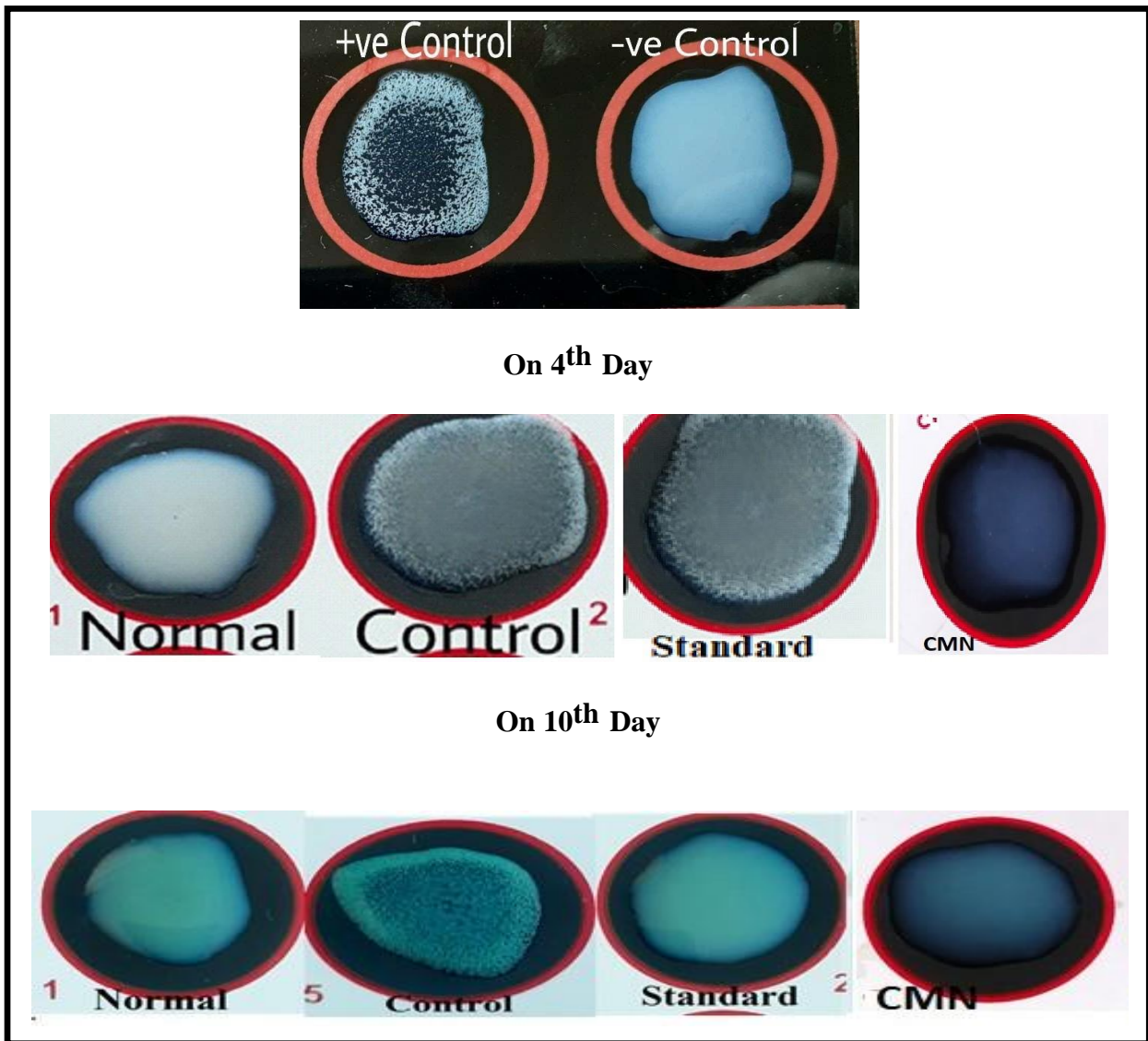


Fig. 11. Effect of formulation dose on C-reactive protein(CRP)

Table 20 Effect of treatment dose on different animal groups.

Groups	Treatment and Dose	Arthritic Index	
		On 4 <sup>th</sup> Day	On 10 <sup>th</sup> Day
1	Normal	Negative	Negative
2	Control	Positive	Negative
3	Standard Diclofenac	Positive	Negative
5	CMN	Positive	Negative

## 4 Conclusion

In this investigation, we used traditional extraction using ethanol and ethyl acetate solvents, TLC detection, and column chromatography technique to identify and isolate certain chemical compounds contained in Commiphora mukul stem. The resulting extract-loaded chitosan nanoparticles were prepared and successfully validated by evaluation parameters. Different composition batches were designed, and herbal tablets were manufactured. *In vitro* drug release suggested that batch F12 is the best amongst all and has good stability over six months. The initial% drug release from the best batch F12 was 92.5, and six months later, it was 91.1. Other factors, such as thickness, weight fluctuation, hardness, and friability, continue to vary little and fall within acceptable limits. Preclinical studies showed that produced nanoparticle formulations responded more favorably than standard and control formulations. Tests on the production formulation's anti-rheumatoid arthritis activity revealed that it had increased therapeutic efficacy and decreased toxicity. This formulation is better for rheumatoid arthritis patients.

## Conflict of interest

No conflict of interest is claimed by the authors.

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## References

- [1] J.S. Smolen, D. Aletaha, I.B. McInnes, Rheumatoid arthritis, Lancet. 388 (2016) 2023–2038. [https://doi.org/10.1016/S0140-6736\(16\)30173-8](https://doi.org/10.1016/S0140-6736(16)30173-8).
- [2] A. Gibofsky, Epidemiology, pathophysiology, and diagnosis of rheumatoid arthritis: A Synopsis., Am. J. Manag. Care. 20 (2014) S128-35. <http://www.ncbi.nlm.nih.gov/pubmed/25180621>.

- [3] L.D. Kumar, R. Karthik, N. Gayathri, T. Sivasudha, Advancement in contemporary diagnostic and therapeutic approaches for rheumatoid arthritis, *Biomed. Pharmacother.* 79 (2016) 52–61. <https://doi.org/10.1016/j.biopha.2016.02.001>.
- [4] A.-F. Radu, S.G. Bungau, Management of Rheumatoid Arthritis: An Overview, *Cells.* 10 (2021) 2857. <https://doi.org/10.3390/cells10112857>.
- [5] P. Prasad, S. Verma, Surbhi, N.K. Ganguly, V. Chaturvedi, S.A. Mittal, Rheumatoid arthritis: advances in treatment strategies, *Mol. Cell. Biochem.* 478 (2023) 69–88. <https://doi.org/10.1007/s11010-022-04492-3>.
- [6] Mickymaray, Efficacy and Mechanism of Traditional Medicinal Plants and Bioactive Compounds against Clinically Important Pathogens, *Antibiotics.* 8 (2019) 257. <https://doi.org/10.3390/antibiotics8040257>.
- [7] Y. Wang, S. Chen, K. Du, C. Liang, S. Wang, E. Owusu Boadi, J. Li, X. Pang, J. He, Y. Chang, Traditional herbal medicine: Therapeutic potential in rheumatoid arthritis, *J. Ethnopharmacol.* 279 (2021) 114368. <https://doi.org/10.1016/j.jep.2021.114368>.
- [8] Y. Huang, Y. Zhao, F. Liu, S. Liu, Nano Traditional Chinese Medicine: Current Progresses and Future Challenges, *Curr. Drug Targets.* 16 (2015) 1548–1562. <https://doi.org/10.2174/1389450116666150309122334>.
- [9] N. Muhamad, T. Plengsuriyakarn, K. Na-Bangchang, Application of active targeting nanoparticle delivery system for chemotherapeutic drugs and traditional/herbal medicines in cancer therapy: a systematic review., *Int. J. Nanomedicine.* 13 (2018) 3921–3935. <https://doi.org/10.2147/IJN.S165210>.
- [10] W.H. De Jong, P.J.A. Borm, Drug delivery and nanoparticles: applications and hazards., *Int. J. Nanomedicine.* 3 (2008) 133–49. <https://doi.org/10.2147/ijn.s596>.
- [11] B. Pelaz, C. Alexiou, R.A. Alvarez-Puebla, F. Alves, A.M. Andrews, S. Ashraf, L.P. Balogh, L. Ballerini, A. Bestetti, C. Brendel, S. Bosi, M. Carril, W.C.W. Chan, C. Chen, X. Chen, X. Chen, Z. Cheng, D. Cui, J. Du, C. Dullin, A. Escudero, N. Feliu, M. Gao, M. George, Y. Gogotsi, A.

Grünweller, Z. Gu, N.J. Halas, N. Hampp, R.K. Hartmann, M.C. Hersam, P. Hunziker, J. Jian, X. Jiang, P. Jungebluth, P. Kadhiresan, K. Kataoka, A. Khademhosseini, J. Kopeček, N.A. Kotov, H.F. Krug, D.S. Lee, C.-M. Lehr, K.W. Leong, X.-J. Liang, M. Ling Lim, L.M. Liz-Marzán, X. Ma, P. Macchiarini, H. Meng, H. Möhwald, P. Mulvaney, A.E. Nel, S. Nie, P. Nordlander, T. Okano, J. Oliveira, T.H. Park, R.M. Penner, M. Prato, V. Puentes, V.M. Rotello, A. Samarakoon, R.E. Schaak, Y. Shen, S. Sjöqvist, A.G. Skirtach, M.G. Soliman, M.M. Stevens, H.-W. Sung, B.Z. Tang, R. Tietze, B.N. Udugama, J.S. VanEpps, T. Weil, P.S. Weiss, I. Willner, Y. Wu, L. Yang, Z. Yue, Q. Zhang, Q. Zhang, X.-E. Zhang, Y. Zhao, X. Zhou, W.J. Parak, Diverse Applications of Nanomedicine., ACS Nano. 11 (2017) 2313–2381. <https://doi.org/10.1021/acsnano.6b06040>.

- [12] S.H. Lee, B.-H. Jun, Silver Nanoparticles: Synthesis and Application for Nanomedicine., Int. J. Mol. Sci. 20 (2019). <https://doi.org/10.3390/ijms20040865>.
- [13] A.S. Barnard, Challenges in modelling nanoparticles for drug delivery., J. Phys. Condens. Matter. 28 (2016) 023002. <https://doi.org/10.1088/0953-8984/28/2/023002>.
- [14] H.-Y. Cho, Y.-B. Lee, Nano-sized drug delivery systems for lymphatic delivery., J. Nanosci. Nanotechnol. 14 (2014) 868–80. <https://doi.org/10.1166/jnn.2014.9122>.
- [15] J. Zhao, Y. Liu, L. Wang, Y. Zhou, J. Du, Y. Wang, Functional and Modified Nanocrystals Technology for Target Drug Delivery., J. Nanosci. Nanotechnol. 18 (2018) 5207–5221. <https://doi.org/10.1166/jnn.2018.15421>.
- [16] M. Amaral, A.B. Pereiro, M.M. Gaspar, C.P. Reis, Recent advances in ionic liquids and nanotechnology for drug delivery., Nanomedicine (Lond). 16 (2021) 63–80. <https://doi.org/10.2217/nnm-2020-0340>.
- [17] B. Mukherjee, Nanosize drug delivery system., Curr. Pharm. Biotechnol. 14 (2013) 1221. <https://doi.org/10.2174/138920101415140804121008>.
- [18] O.C. Farokhzad, R. Langer, Impact of nanotechnology on drug delivery., ACS Nano. 3 (2009) 16–20. <https://doi.org/10.1021/nn900002m>.

- [19] C. Jin, K. Wang, A. Oppong-Gyebi, J. Hu, Application of Nanotechnology in Cancer Diagnosis and Therapy - A Mini-Review., *Int. J. Med. Sci.* 17 (2020) 2964–2973. <https://doi.org/10.7150/ijms.49801>.
- [20] Y. Zhang, M. Li, X. Gao, Y. Chen, T. Liu, Nanotechnology in cancer diagnosis: progress, challenges and opportunities., *J. Hematol. Oncol.* 12 (2019) 137. <https://doi.org/10.1186/s13045-019-0833-3>.
- [21] P. Mulvaney, Nanoscience vs nanotechnology--defining the field., *ACS Nano.* 9 (2015) 2215–7. <https://doi.org/10.1021/acsnano.5b01418>.
- [22] P.S. Weiss, Nanoscience and nanotechnology: present and future., *ACS Nano.* 4 (2010) 1771–2. <https://doi.org/10.1021/nn100710n>.
- [23] D. Bonnell, The next decade of nanoscience and nanotechnology., *ACS Nano.* 4 (2010) 6293–4. <https://doi.org/10.1021/nn102952y>.
- [24] E.S. Ong, Extraction methods and chemical standardization of botanicals and herbal preparations., *J. Chromatogr. B. Analyt. Technol. Biomed. Life Sci.* 812 (2004) 23–33. <https://doi.org/10.1016/j.jchromb.2004.07.041>.
- [25] V. Garg, V.J. Dhar, A. Sharma, R. Dutt, Facts about standardization of herbal medicine: a review., *Zhong Xi Yi Jie He Xue Bao.* 10 (2012) 1077–83. <https://doi.org/10.3736/jcim20121002>.
- [26] J. Prakash, S. Srivastava, R.S. Ray, N. Singh, R. Rajpali, G.N. Singh, Current Status of Herbal Drug Standards in the Indian Pharmacopoeia., *Phyther. Res.* 31 (2017) 1817–1823. <https://doi.org/10.1002/ptr.5933>.
- [27] W. World Health Organization, Quality control methods for medicinal plant materials World Health Organization Geneva, (1998). [https://who.int/docs/default-source/medicines/norms-and-standards/guidelines/quality-control/quality-control-methods-for-medicinal-plant-materials.pdf?sfvrsn=b451e7c6\\_0](https://who.int/docs/default-source/medicines/norms-and-standards/guidelines/quality-control/quality-control-methods-for-medicinal-plant-materials.pdf?sfvrsn=b451e7c6_0) (accessed July 7, 2023).