



FORMULATION AND IN VITRO EVALUATION OF CIPROFLOXACIN SUSTAIN RELEASE TABLETS

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Abstract:

Ciprofloxacin has a short biological half-life of 4–6 hours and having less bioavailability, which necessitates multiple daily dosing. Hence, the present study was aimed to develop a SUSTAIN release formulation of Ciprofloxacin to reduce the dose related side effects and to reduce the dosage regimen. The present research project aimed to develop a Control release oral formulation of hypertension drug Ciprofloxacin, the present research comprising Ciprofloxacin used for the symptomatic relief of pulmonary arterial hypertension. Polymers like Xanthan Gum, Carbopol 940, and Ethyl Cellulose were used for controlling the drug release, and the polymers are mixed in a predetermined ratio. Totally 12 formulations were prepared and evaluated for pre-compression and post-compression parameters, and all the results were found to be within the limits. From the drug and excipients compatibility studies (FT-IR) it was confirmed that the drug and excipients used weren't have any interactions. The in vitro dissolution studies revealed that the F8 formulation containing 250mg of Carbopol 940 controls the drug release upto 12 hours. So Carbopol 940 containing F8 formulation was considered to be suitable for the formulation of Ciprofloxacin SUSTAIN release tablet and the drug release kinetics revealed that the F8 formulation shows zero order kinetics with super case- II transport mechanism.

Keywords: Ciprofloxacin, Carbopol 940, sustain release, FT-IR.

1. INTRODUCTION

Sustained release dosage forms are the substances which initially release drug sufficiently to provide therapeutic effect soon after administration and further a gradual release over an extended period of time³. Not all drug candidates are suitable for the preparation of sustained release dosage forms. The understanding of the certain parameters are very essential in the development of sustained release dosage forms, those are mainly drug physico-chemical characteristics such as, aqueous solubility, partition coefficient, ionization constant, drug stability⁴ and biological parameters viz., absorption, distribution, metabolism, excretion, biological half life, therapeutics of drug in vivo environment⁵ and also dose size and dosage frequency. It is always been a challenging task to the pharmaceutical scientist to prepare suitable sustained release dosage forms in pharmaceutical industry, because there is a possibility of dose dumping due to food, poor in vitro - in vivo correlation, decreased systemic availability in comparison with conventional dosage forms due to incomplete release or increased first pass metabolism⁶. Being several types of solid oral dosage forms, tablets offer more flexibility in the preparation to scientists when compare to other liquid and semi solid dosage forms. The various approaches to prepare essential solid oral sustained release drug delivery systems are given summarized below¹⁻⁴

I. Diffusion-controlled DDS

- Hydrophobic matrix systems
- Hydrophilic matrix systems
- Inert matrix systems
- Reservoir matrix systems

II. Dissolution - controlled DDS

- Based on dissolution-controlled release of solid particles
- Based on dissolution-controlled release coated technologies
- Based on dissolution-controlled release matrix technologies

III. Osmotic controlled DDS

IV. Biodegradable polymeric DDS

- Microparticles
- Ion Exchange resins
- Nanoparticles
- V. Ligand-based targeting DDS
- VI. Stimulus responsive DDS
- Physically modulated: Temperature
- Chemically modulated: pH dependent

Although the above systems are commonly used for sustaining the drug delivery, there are some practical limitations of these techniques. The applicability of osmotic drug delivery technology for sustaining the drug release is very less due to the preparation aspects, quality control of these systems are more expensive when compare to other drug delivery systems and also getting the integrity and consistency of coating is very hard. The rate of production is the main limitation for microspheres and high dose is one of the major limitations for preparation of nanoparticles. Ligand based systems are especially used as target drug delivery systems to a specific site. Stimulus responsive drug delivery systems release the medicament depending up on the biological rhythms and this technology is not suitable to all categories of drugs. The drugs which are having high absorption window in the upper part of the gastric region are only suitable for gastric floating systems. Among all these drug delivery systems just mentioned, especially diffusion and dissolution controlled drug delivery systems such as hydrophilic and hydrophobic matrix systems are very interesting and suitable for most categories of drugs.

Matrix Systems: Matrix devices consist of drug dispersed homogeneously throughout a continuous phase of polymer or lipid. The device can be prepared either by the compression of polymer/drug mixture. Preparation of matrix tablets are widely been used now a days due to least complicated approach for retarding the release rate of drugs over 10 an extended period of time⁵⁻⁸ because their formulation is simpler, inexpensive, easy to produce¹³ and they have good in vitro-in vivo correlation. A matrix system is capable of accommodating both low and high drug loading active ingredients with a wide range of physical and chemical properties. Based up on types of release retardants the matrix systems are divided into two categories such a. Hydrophilic matrix systems b. Hydrophobic matrix systems

1.1. Extended-Release Dosage Forms

Dosage forms which can reduce at least a twofold reduction in dosing frequency as compared to the drug presented in a conventional form, such as solution or a prompt-releasing conventional solid dosage form are termed as extended release dosage forms.

These products are formulated to make the contained medicament available over an extended period of time after administration within its therapeutic range and hence reduction in dosing frequency as compared to the conventional dosage forms.⁹⁻¹²

They include:

- **Controlled Release (Ideal Zero-Order)**
- **Prolonged Release**

Comparison of different types of modified release dosage formulations as per plasma concentration vs. time can be explained by the following figure.

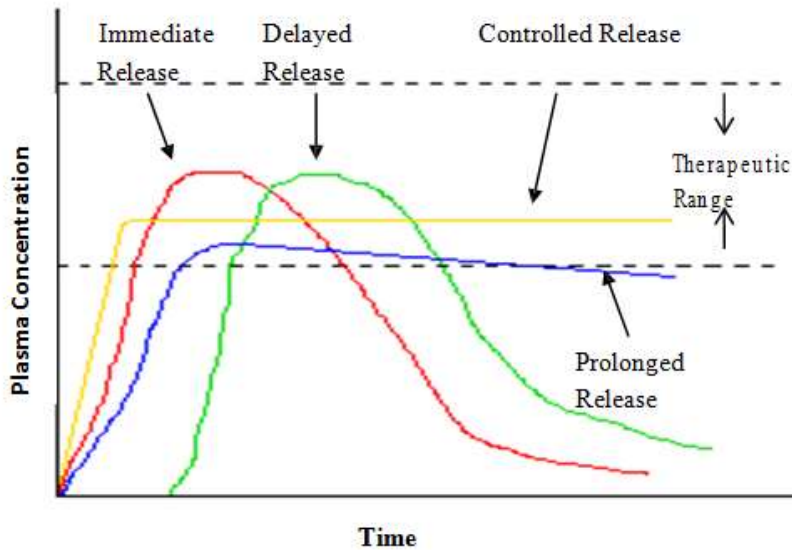


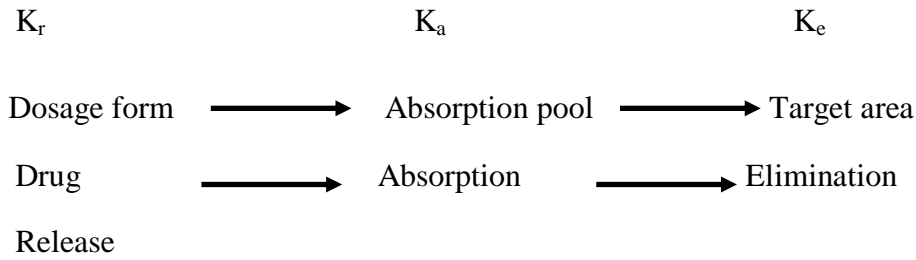
Fig1: Plasma concentration vs. Time curve for different type of dosage forms

Modified-release products can also be explained by other different terms such as extended-release, prolonged-release, controlled-release, controlled-delivery, slow-release and sustained-release, delayed release, time release etc. These formulations, by definition, have a reduced rate of release of active substance. In general, these terms are interchangeable¹³.

A delayed-release product is a modified-release, but by definition is not an extended-release. They can release discrete amount(s) of drug some time after drug administration, e.g. enteric-coated products exhibit a lag time during which little or no absorption occurs. While a number of such modified-release products are available as both prescription and over-the-counter drugs, only a limited number have been shown to offer a therapeutic advantage¹³⁻¹⁴.

Release Rate and Dose Consideration¹⁵

Conventional dosage forms include solutions, capsules, tablets, emulsions, etc. These dosage forms can be considered to release their active ingredients into an absorption pool immediately.



Where,

K_r = First order rate constant for drug release.

K_a = First order rate constant for drug absorption.

K_e = First order rate constant for overall drug elimination.

□ For immediate release dosage forms $K_r \gg K_a$ or alternatively absorption of drug across a biological membrane is the rate-limiting step in delivery of the drug to its target area.

□ For non-immediate release dosage forms, $K_r \ll K_a$, that is, release of drug from the dosage form is the rate-limiting step. This causes the above kinetics scheme to reduce to

Thus, the effort to develop a delivery system that releases drugs slowly must be directed primarily at altering the release rate by affecting the value of K_r . The ideal goal in designing a controlled-release system is to deliver drug to the desired site at a rate according to needs of the body, i.e. a self-regulated system based on feedback control but this is a difficult assignment.

Terminology¹⁶:

Controlled drug delivery or modified release delivery systems may be defined as follows: -

Controlled– Release formulation:

The controlled release system is to deliver a constant supply of the active ingredient, usually at a zero-order rate, by continuously releasing, for a certain period, an amount of the drug equivalent to the eliminated by the body. An ideal controlled drug delivery is the one, which delivers the drugs at a predetermined rate, locally or systemically, for a specific period of time.

Repeat action preparations:

A dose of the drug initially is released immediately after administration, which is usually equivalent to a single dose of the conventional drug formulation. After a certain period, a second single dose is released. In some preparation, a third single dose is released after a certain time has elapsed, following the second dose. The main advantage is that it provides the convenience of supplying additional dose(s) without the need of re-administration. It has a disadvantage that the blood levels still exhibit the “Peak and valley” characteristic of conventional intermittent drug therapy.

Extended-Release formulations:

Extended-Release formulations are usually designed to reduce dose frequency and maintain relatively constant or flat plasma drug concentration. This helps avoid the side effects associated with high concentration.

Delayed release preparations:

The drug is released later after administration. The delayed action is achieved by the incorporation of a special coat, such as enteric coating, or other time barriers such as the formaldehyde treatment of soft and hard gelatin capsules. The purposes of such preparations are to prevent side effects related to the drug presence in the stomach, protect the drug from degradation in the highly acidic H^+ of the gastric fluid.

Sitespecific targeting:

These systems refer to targeting of a drug directly to a certain biological location. In this case the target is adjacent to or in the diseased organ or tissue.

Receptor targeting:

These systems refer to targeting of a drug directly to a certain biological location. In this case the target is the particular receptor for a drug within an organ or tissue. Sitespecific targeting and receptor targeting systems satisfy the spatial aspect of drug delivery and are also considered to be controlled drug delivery systems.

1.2 ADVANTAGES AND DISADVANTAGES OF ORAL EXTENDED-RELEASE DOSAGE FORMS

All controlled release dosage forms have a common goal of improving the drug therapy compared to that achieved by their non sustained counterparts.

Advantages

- Avoid patient compliance problems
- Employ less quantity drug.
- Minimize or eliminate local side effects.
- Minimize or eliminate systemic side effects.
- Reduced dosing frequency and fluctuation of therapeutic plasma concentration
- Obtain less potentiation or reduction in drug activity with chronic use.
- Minimized drug accumulation with chronic dosing.
- Improves efficiency in treatment
- Cures or control conditions more promptly.
- Improves control of condition i.e., reduces fluctuations in drug level.
- Improves bioavailability of some drugs.
- Makes use of special effects in sustained release aspirin for morning relief of Arthritis by dosing before bedtime.

Disadvantages

- Dosedumping, toxicitycan occurif system fails.
- Reduced potentina patient education.
- Does not permit theprompttermination ofdrugtherapy.
- Cannot exercise anycontrol oncethe dosage formis administered.

1.3 ORAL CONTROLLED– RELEASEPRODUCTS:

Based on the releasemechanismthese are classified asfollows¹⁷:-

1. Diffusion-controlled products.
2. Dissolution-controlled products.
3. Erosion products.
4. Osmotic pump systems.
5. Ionexchangeresins.

1. Diffusion– Controlled products:

Inthese systems,there iswater–insoluble polymer whichcontrolsthe flow ofwater andthe subsequent release ofdissolved drugfromthedosage form. Diffusionoccurswhenadrug passsthroughthepolymerthatformsthecontrolled release device.The diffusioncanoccur throughporesinthe polymermatrixor by passingbetween polymerchains. Thesearebroadlydivided into two categories:-

A. ReservoirDevices.

B. MatrixDevices.

Thebasicmechanismsofdrugreleasefromthesetwosystemsarefundamentally different.

A. Reservoir Devices

Inthissystema water insoluble polymeric materialencasesa core of drug.Drug willpartitionintothemembraneandexchange withthefluidsurrounding the particles (or)tablet.

Theactiveagentisreleasedtothesurrounding environmentby diffusionprocessthroughtheratelimitingmembrane.Inthereservoirsystemsthe drugdeliveryrate remains fairlyconstant.

B. Matrix Devices

Inthematrixdevicesthedrug oractiveisdispersedinpolymermatrix toforma homogeneousystemknownasa matrixsystem.Diffusionoccurswhenthe drugpassesfromthepolymer matrixintotheexternalenvironment.Asthe release continues,itsratenormally decreaseswiththistypeofsystem,sincetheactiveagent hasaprogressively longerdistancetotravelandthereforerequiresalongerdiffusion time to release.

2. Dissolution-controlled products:

In these products, the rate of dissolution of the drug is controlled by slowly soluble polymers or by microencapsulation. Once the coating is dissolved, the drug becomes available for dissolution. By varying the thicknesses of the coat and its composition, the rate of drug release can be controlled. Some preparations contain a fraction of the total dose as an immediate-release component to provide a pulsed dose soon after administration. The pellet

dosage forms of diffusion- or dissolution-controlled products can be encapsulated or prepared as a tablet.

Dissolution-controlled products can be sub-divided into two types:-

- A. Encapsulation Dissolution controls.
- B. Matrix Dissolution control.

A. Encapsulation Dissolution control

This system method involves coating of individual particles (or) granules of drug with a slow dissolving material. The coated particles can be compressed directly into tablets (or) placed in capsules. The rate of dissolution of the drug (and thereby availability for absorption) is controlled by microencapsulation. Once the coating is dissolved, the drug becomes available for dissolution. By varying the thicknesses of the coat and its composition, the rate of drug release can be controlled.

These products should not be chewed as the coating may be damaged. One of the advantages of encapsulated pelleted products is that the onset of absorption is less sensitive to stomach emptying.

The entrance of the pellets into the small intestine (where the majority of drug absorption occurs) is usually more uniform than with non-disintegrating sustained-release tablet formulations.

B. Matrix Dissolution control

In this system an alternative approach is to compress the drug with a slow dissolving carrier. Here the rate of drug release is controlled by the rate of penetration of the dissolution fluid into the matrix, porosity, presence of hydrophobic additives and the wet ability of system and surface of particle.

3. Erosion products:

In this system drug or active agents are mixed with biodegradable polymers. These materials degrade within the body as a result of natural biological processes and drug release occurs at a constant rate.

Most biodegradable polymers are designed to degrade as a result of hydrolysis of the polymer chains into biologically acceptable and progressively smaller compounds. The release of drug from these products is controlled by the erosion rate of a carrier matrix. The rate of release is determined by the rate of erosion.

4. Osmotic pump systems:

The osmotic pump is similar to a reservoir device but contains an osmotic agent (e.g., the active agent in salt form) which acts to imbibe water from the surrounding medium via a semi-permeable membrane.

Pressure is generated within the device which forces the active agent out of the device via an orifice (of a size designed to minimize solute diffusion, whilst preventing the build-up of a hydrostatic pressure head which has the effect of decreasing the osmotic pressure and changing

the dimensions {volume} of the device). The advantage of this type of product is that the constant release is unaltered by the environment of the gastrointestinal tract and relies simply on the passage of water into the dosage form.

The rate of release can be modified by altering the osmotic agent and the size of the hole.

5. Ion exchange resins:

Drug-resin complexes ("resonates") for extended release are known and have been successfully used commercially. The drug is bound to the resin and released by exchanging with appropriately charged ions in contact with the ion exchange groups. This technique is applicable to certain drugs which have particular characteristics in terms of their relative affinity for the polymers being used.

1.4 Drug Properties Relevant To Controlled- Release Formulations

The extent of fluctuation in drug concentration at steady state is determined by the relative magnitude of the elimination half-life and the dosing interval. If a drug is given at an interval equal to the elimination half-life, there is a two-fold difference between the maximum and minimum concentrations at steady state.

Some drugs that have relatively high solubility at the low pH with short biological half-life, are not suitable for conventional oral dosage formulations, because the high acid solubility property of drug results in rapid drug absorption and clearance, causing large and undesirable fluctuations in plasma concentration.

For drugs with short half-lives and with a clear relationship between concentration and response, it will be necessary to dose at regular, frequent intervals in order to maintain the concentration within the therapeutic range. Higher doses at less frequent intervals will result in higher peak concentrations with the possibility of toxicity. For some drugs with wide margins of safety, this approach may be satisfactory, e.g. amoxicillin has a half-life of approximately one hour, but a dosage frequency of 8 hours.

Judicious choice of the drug substance is the most important decision in the successful development of controlled release product. Several categories of drug have potential for their therapeutic improvement of efficacy via controlled-release oral routes e.g. antianginal, Anti-inflammatory, Antihistaminic, Antigastric resistant agents, Antipsychotic agents, Antidiabetic drug of agents. The common goal for increased duration is twice a day, or when feasible, once a day. Several properties of the drug itself can lead to the achievement of a 12 to 24 hour oral prolonged release dosage form. Some of the characteristics mitigating against success are the following:-

1. Very short half-life and/or a relatively large single dose.
2. Long half-life.
3. Potent drug with a low margin safety.
4. Poorly soluble and/or poorly absorbed.
5. Biological activity not a function of core in blend.
6. Absorption primarily active through a "window".
7. Large first-pass metabolism.

The selection of both the drug and retardant polymers along with the filler excipients will impact on the mechanism and rates of drug release from the dosage formulation. Various physicochemical, biological properties of a drug and its biopharmaceutical characteristics influencing in product design and performance.

A. Physicochemical properties

1. Aqueous solubility

A drug with good aqueous solubility serves as a good candidate for controlled release dosage forms, e.g., pentoxifylline. Compounds with very low aqueous solubility usually suffer from bioavailability problems because of limited gastrointestinal transit time of the undissolved drug particles and limited solubility at the absorption site.

2. Partition coefficient

Between the time that a drug is administered and the time it is eliminated from the body, it must diffuse through a variety of biological membranes with a primary barrier as a lipid-like barrier. A major criterion in the evaluation of the ability of a drug to penetrate these lipid membranes is its apparent oil/water partition coefficient, defined as

$$K = C_o / C_w$$

Where

C_o = total concentration of all forms of the drug, e.g., ionized and unionized, in some organic phase at equilibrium, and

C_w = total concentration of all forms in an aqueous phase at equilibrium.

Drugs with a partition coefficient that is either extremely higher or lower than the optimum are, in general, poorer candidates for formulation into modified-release dosage forms.

3. Drug Stability

Stability is another physicochemical property to be considered in the design of sustained-release systems. Drugs that are unstable in the stomach release their contents only in the intestine. So it can be replaced in a slowly soluble form.

Stability studies are extremely important so that the pharmaceutical manufacturer can accurately predict the shelf-stability of a new product from accelerated storage stability data. Drugs with significant stability problems in any particular area of the gastrointestinal tract are less suitable for formulation into controlled release systems.

4. Protein binding

The drug-protein complex can serve as a depot for producing a prolonged release profile, drug that exhibits a high degree of binding to plasma. The main forces of attraction responsible for binding are van der Waals forces, hydrogen bonding, and electrostatic forces. Drugs bound to mucin may increase absorption. e.g.: Quaternary ammonium compounds bound to mucin in the gastrointestinal tract.

5. Molecular Size and diffusivity

In addition to diffusion through these biological membranes, in many sustained-release systems must diffuse through a polymeric membrane or matrix that is used to control their release kinetics. The ability of a drug to diffuse through a polymeric membrane or matrix that is used to control their release kinetics is a function of its diffusivity (diffusion

coefficient).

An important influence upon the value of the diffusivity (D) in polymers is the molecular size (or molecular weight) of the diffusing species. In most polymers, it is possible to relate $\log D$ empirically to some function of molecular size. The value of D thus is related to the size and shape of the cavities as well as size and shape of drugs.

Generally, the values of the diffusion coefficient for intermediate molecular weight drugs, i.e. 150-400, through flexible polymers range from 10^{-6} to 10^{-9} cm²/sec, with values on the order of 10^{-8} being most common.

B. Biological Factors

1. Absorption

The rate, extent, and uniformity of absorption of a drug are important factors when considering its formulation into a controlled release system. Since the rate limiting step in drug delivery from a controlled release system is its release from a dosage form, rather than absorption, a rapid rate of absorption of the drug relative to its release is essential if the system is to be successful. To maintain constant blood

(or) tissue level of drug it must be uniformly released from the controlled release system, and then uniformly absorbed. Design of controlled release product would be more difficult with respect to the oral route. e.g.: Quaternary ammonium compounds; amino glycosides such as Gentamycin.

2. Distribution

Distribution of a drug into vascular and extravascular spaces in the body is an important factor in its overall drug elimination kinetics. Two parameters that are used to describe the distribution characteristics. The term apparent volume of distribution of a drug is frequently used to describe the magnitude of distribution. In the case of two compartment model the total apparent volume of distribution for a drug at steady state can be calculated from the following equation.

Where,

$$V_{d_{ss}} = [(K_{12} + K_{21}) / K_{21}] V_p$$

$V_{d_{ss}}$ = Apparent volume of distribution at steady state.

K_{12} = Constant for central to peripheral compartment.

K_{21} = Constant from peripheral to central compartment.

V_p = Volume of central compartment

3. Metabolism

The metabolic conversion of a drug to another chemical from usually can be considered in the design of a controlled-release system for that drug. There are two factors associated with metabolism of some drugs. One is ability of the drug to induce (or) inhibit enzyme synthesis and the other is fluctuating drug blood level due to hepatic first pass effect. For example the organ most responsible for metabolism is the liver. Metabolism of a drug will be reflected in the elimination constant of a drug.

4. Elimination and biological half-life

The rate of elimination of a drug is described quantitatively by its biological half-life of a drug is related to its apparent volume of distribution (V) and its systemic clearance.

$$t_{1/2} = 0.693 V / Cl_s = 0.693 V AUC / \text{dose}$$

Where,

Cl_s = systemic clearance.

A drug with a short half-life requires frequent dosing, and this makes it a desirable candidate for a controlled-release formulation. On the other hand, a drug with a long half-life is dosed at greater time intervals, and thus there is a need for a controlled-release system.

5. Side effects

Most of the drugs will produce side effects. For some drugs, the incidence of side effects is believed to be a function of plasma concentration, and it can be minimized by controlling the concentration at which the drug exists in plasma, and hence controlled release formulations appear to offer a solution to this problem.

6. Margin of safety

The margin of safety of a drug is mostly used to measure its therapeutic index. In the case of controlled release, one can consider to be therapeutically safe and effective in drug therapy monitoring. Especially for potent drugs, whose therapeutic concentration range is narrow, the value of therapeutic index is small. **e.g.:** Cardiac glycosides, Anti-arrhythmic drugs etc.

7. Total clearance

Clearance is defined as the theoretical volume of body fluid containing drug, from which the drug is completely removed in a given period of time, it is expressed in ml/min (or) liters / hour. Clearance is given by,

Rate of elimination

$$\text{Total Clearance} = \frac{\text{Rate of elimination}}{\text{Plasma drug concentration}}$$

8. Mean residence time (MRT)

Mean residence time is defined as the time of drug molecule residence in the body. It is the time corresponding to 63.2% elimination from the body. It is calculated from AUC and AUMC.

$$\text{MRT} = \text{AUMC} / \text{AUC}$$

Where,

MRT = Mean residence time.

AUMC = Area under the first-moment curve. AUC = Area under the zero-moment curve.

1.5. Matrix Technology

Matrix technologies are popularly used because of the simplicity of the manufacturing processes required, level of reproducibility, stability of the raw materials and dosage forms as well as ease of scale up operation, validation and favorable in-vitro in-vivo correlation (IVIVC). Classically, simple matrix delivery systems exhibit first order or square root of time release kinetics.

These systems improve patient compliance and decrease incidence of adverse drug reactions. Under ideal conditions, a controlled-release formulation maintains therapeutic blood level of a drug for a specific period of time. A number of oral controlled-release dosage forms have been developed and studied to restrict these systems to specific regions of the gastrointestinal tract as well as to improve the pharmacological activity and to reduce toxic effects. In order to overcome all those problems mentioned above, the matrix tablet has additional advantage like, Matrix tablets are resistant to dose dumping. They are simple in nature of the formulations, and due to robustness they are unaffected by variations in ingredients^{12,18}.

Matrix tablets containing hydrophilic polymers are a common and commercially successful means of prolonging oral drug delivery. A common problem observed with hydrophilic matrix systems containing water soluble drugs is an initial burst effect of the drug release^{12,18}.

Process of Manufacturing Matrix Tablets

One of the commonly employed processes for the manufacture of extended release dosage forms involves the direct compression of blends of drug, retardant material, and additives to form a tablet in which the drug is embedded in the matrix core of retardant. Alternately, the retardant-drug blends can be granulated prior to compression.

Matrix devices are of two types: Matrix dissolution controlled and matrix diffusion controlled drug delivery devices¹⁸.

Matrix Diffusion Controlled Drug Delivery Devices.

Matrix diffusion devices are prepared by dispersing a solid drug in an insoluble polymer matrix carrier system, i.e. a drug reservoir is formed by homogenous

dispersion of solid drug particles throughout a lipophilic or hydrophilic polymer matrix. The rate of drug release is dependent on the rate of drug diffusion but not on the rate of solid dissolution. The equation describing drug from this system T. Hugiichi has derived this system¹⁸.

$$Q = C_s \sqrt{\frac{D \varepsilon (2A - \varepsilon C_s)}{\tau}}$$

Where,

Q = Weight in grams of drugs in the unit surface area.

D = Diffusion coefficient of drug in the release medium.

ε = Porosity of the matrix.

τ = Tortuosity of the matrix.

C_s = Solubility of the drug in release medium and,

A = Concentration of drug in the tablet expressed as g/ml.

The following assumptions were made in deriving the above equation: A pseudo-steady state is maintained during release.

$A \gg C_s$ i.e. excess solute is present.

$C = 0$ in solution at all times (perfect sink condition) Drug particles are much smaller than those in the matrix. The diffusion coefficient remains constant.

No interactions between the drug and the matrix occur.

On many controlled drug release from a homogeneous matrix by varying the following parameters.

Initial concentration of drug in the matrix.

Drug solubility.

Porosity

Tortuosity

Leaching solvent composition.

Polymer system making up matrix.

Matrix Dissolution Controlled Drug Delivery Devices

Matrix dissolution devices can also be formulated by compressing the drug with a slowly dissolving polymer carrier into a tablet form. There are two general methods of preparing drug-wax particles: congealing method and aqueous dispersion method. In the congealing method, the drug is mixed with wax material and either spray congealed or congealed and screened. In the aqueous dispersion method, the drug-wax mixture is sprayed or placed in water and the resultant particles are collected.

Matrix tablets are made by direct compression of a mixture of drug, polymer and excipients. The rate-limiting step in controlling release from these formulations is liquid penetration into the matrix. Some channeling agents (wetting agents) can be incorporated with the blend of mixture to promote permeation of polymer matrix by water, which allows drug dissolution and diffusion from the channels created in the matrix. Formulations should be designed, so that pore diffusion becomes the rate-controlling step.

Drug bioavailability, which is critically dependent on the drug:polymer ratio, may be modified by inclusion of diluents such as lactose in place of polymer in low-milligram-potency formulations.

Drug release is controlled by penetration of water through a gelled layer produced by hydration of polymer and diffusion of drug through swollen, hydrated matrix, in addition to erosion of the gelled layer. The polymer selected for formulation as well as the drug:polymer ratio control the extent to which diffusion or erosion which controls release of the drug from the formulation¹⁸.

- Optimization of formulation parameters and drug-carrier system using appropriate methods.
- Percentage of drug content.
- Study of Dissolution
- Release kinetic studies
- Drug interaction studies by FTIR.

METHODOLOGY**Materials used:**

S. No.	INGREDIENTS AND AGENTS	MANUFACTURER / SUPPLIERS
1.	Ciprofloxacin	A. R. Life Sciences, Hyderabad
2.	Xanthan Gum	Strides arcolab, Bangalore.
3.	Carbopol 940	Strides arcolab, Bangalore.
4.	Ethyl Cellulose	Lobachemiepvt.ltd, Mumbai
5.	MCC	Lobachemiepvt.ltd, Mumbai
6.	Lactose	Lobachemiepvt.ltd, Mumbai
7.	PVP K 30	Lobachemiepvt.ltd, Mumbai
8.	Magnesium Stearate	Lobachemiepvt.ltd, Mumbai
9.	Talc	Lobachemiepvt.ltd, Mumbai

Instruments used:

Sr. No.	Instruments	Manufacturer
1	UV-Vis Spectrophotometer	T-60 UV-Visible Spectrophotometer.
2	FTIR 1700S Spectrophotometer	Shimadzu, Japan.
3	Electronic Weighing Balance	Sartorius Ltd.
	Dissolution test apparatus TDT-08T Dissolution Tester (USP)	LAB India DS-8000
5	Digital Vernier Caliper	Dgisun Electronics Hyderabad.
6	Digital pH meter 7007	Dgisun Electronics Hyderabad.
7	Test Sieve (No. 40, 60)	Scientific Engineering Corp. Delhi.
8	Hot Air Oven	Servewell Instrument PVT LTD, Bangalore.
9	Stability Chamber	Lab Control Equipment Co. Mumbai.
11	Friabilator USPEF-2	Eleectrolab, Mumbai
12	Tablet punching machine (Rimek mini press-1) (10 stations 9.25 mm concave punches)	Karnavati Engineering Ltd, Mehsana, Gujarat.
13	Monsanto Hardness Tester	Ketan engineering Ltd, Mumbai

Pre formulation studies⁴²⁻⁴³

Pre formulation testing is an investigation of physical and chemical properties of drug substances alone and when combined with pharmaceutical excipients. It is the first step in the ratio development of dosage form.

Solubility

Solubility of Ciprofloxacin was determined in pH 1.2 and pH 7.4 and 6.8 phosphate buffers.

Solubility studies were performed by taking excess amount of Ciprofloxacin in beakers containing the solvents. The mixtures were shaken for 24 hrs at regular intervals. The solutions were filtered by using Whatmann's filter paper grade no. 41. The filtered solutions are analyzed by spectrophotometrically.

Melting point:

Melting point of the drug was determined by taking small amount of drug in a capillary tube closed at one end. The capillary tube was placed in an electrically operated melting point apparatus and the temperature at which the drug melts was recorded. This was performed thrice and average value was noted.

Compatibility Studies

Compatibility study with excipients was carried out by FTIR. The pure drug and its formulations along with excipients were subjected to FTIR studies. In the present study, the potassium bromide disc (pellet) method was employed.

Identification of Ciprofloxacin⁴⁴

PREPARATION OF BUFFERS

A. 0.1 N HCL

Dilute 8.50 ml of HCL with distilled water to make up the volume to 1000 ml

B. 6.8 pH Phosphate buffer

Dissolve 13.872 g of potassium dihydrogen phosphate and 35.084 g of disodium hydrogen phosphate in sufficient water to produce 1000 ml.

C. 7.4 pH Phosphate buffer

Dissolve 0.6 g of potassium dihydrogen phosphate and 6.4 g of disodium hydrogen phosphate and 5.85 g of sodium chloride in distilled water and dilute up to 1000.0 ml.

Determination of UV spectrum of Ciprofloxacin:

10 mg of Ciprofloxacin was dissolved in 10 ml of buffer, so as to get a stock solution of 1000 µg/ml concentration. From the above stock solution pipette out 1 ml of the solution and make up the volume to 10 ml using buffer to get the concentration of 100 µg/ml concentration. From this stock solution pipette out 1 ml of the solution and make up the volume to 10 ml using buffer to get the concentration of 10 µg/ml concentration, this solution was scanned under UV Spectroscopy using 200-400 nm.

Preparation of Standard Calibration Curve of Ciprofloxacin in pH 1.2:

A. Preparation of Stock Solution

10 mg of Ciprofloxacin was dissolved in 10 ml of pH 1.2 buffers so as to get a stock solution of 1000 µg/ml concentration.

B. preparation Standard Solution

1 ml of stock solution was diluted to 10 ml with pH 1.2 buffer in 10 ml volumetric flask this gives a concentration of 100 µg/ml. Aliquot of standard drug solutions were prepared by withdrawing 0.2, 0.4, 0.6, 0.8, 1.0 and 1.2 ml and transferred in to 10 ml volumetric flask and were diluted up to the mark with pH 1.2 buffer. This gives the final concentration of 2, 4, 6, 8, 10 and 12 µg/ml of Ciprofloxacin respectively. The absorbances of the solution were measured against pH 1.2 as blank at 279 nm using UV visible spectrophotometer. The absorbance values were plotted against concentration (µg/ml) to obtain the standard calibration curve.

Preparation of Standard Calibration Curve of Ciprofloxacin in pH 6.8:

A. Preparation of Stock Solution

10mg of Ciprofloxacin was dissolved in 10ml of pH 6.8 phosphate buffer so as to get a stock solution of 1000µg/ml concentration.

B. preparation Standard Solution

1ml of stock solution was diluted to 10ml with pH 6.8 buffer in 10ml volumetric flask this gives a concentration of 10µg/ml. Aliquot of standard drug solutions were prepared by withdrawing 0.2, 0.4, 0.6, 0.8, 1.0 and 1.2ml and transferred in to 10ml volumetric flask and were diluted up to the mark with pH 6.8 buffer. This gives the final concentration of 2, 4, 6, 8, 10 and 12µg/ml of Ciprofloxacin respectively. The absorbances of the solution were measured against pH 6.8 as blank at 279 nm using UV visible spectrophotometer. The absorbance values were plotted against concentration (µg/ml) to obtain the standard calibration curve.

Preparation of Ciprofloxacin Controlled Release Tablets:⁴⁴⁻⁴⁸

Controlled release tablets of Ciprofloxacin were prepared by Direct compression method using variable concentrations of different polymers like Xanthan Gum, Carbopol 940, Guar Gum. Direct compression method is widely employed method for production of compressed tablets.

Direct Compression:

In this process the tablets are compressed directly from powder blends of active ingredient and suitable excipients, which will flow uniformly in to the die cavity and forms a firm compact.

Brief manufacturing procedure for the preparation of tablets:

Step 1- Weighed all the ingredients separately.

Step 2- The drug and the other excipients were passed through 40# sieve together and blended for 10 minutes.

Step 3- The magnesium stearate was passed through 60# sieve and added to the blend of step 2 and blended for 5 minutes.

Step 4- Compressed the blend of step 3 in to tablets by using 8.5mm, round punches.

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Tablet composition of different formulations of Ciprofloxacin Controlled Release tablets

Ingredients(mg)	F1	F2	F3	F4	F5	F6	F7	F8	F9	F10	F11	F12
Ciprofloxacin	250	250	250	250	250	250	250	250	250	250	250	250
MCC	216	166	116	66	216	166	116	66	216	166	116	66
Xanthan Gum	100	150	200	250	-	-	-	-	-	-	-	-
Carbopol 940	-	-	-	-	100	150	200	250	-	-	-	-
Ethyl cellulose	-	-	-	-	-	-	-	-	100	150	200	250
Lactose	10	10	10	10	10	10	10	10	10	10	10	10
PVP K30	20	20	20	20	20	20	20	20	20	20	20	20
Talc	2	2	2	2	2	2	2	2	2	2	2	2
Mg. stearate	2	2	2	2	2	2	2	2	2	2	2	2
Total (mg)	600	600	600	600	600	600	600	600	600	600	600	600

Evaluation Parameters ⁴⁹⁻⁵²**Pre Compression Parameters****A. Bulk density (D_b)**

It is the ratio of powder to bulk volume. The bulk density depends on particle size distribution, shape and cohesiveness of particles. Accurately weighed quantity of powder was carefully poured into graduated measuring cylinder through large funnel and volume was measured which is called initial bulk volume. Bulk density is expressed in gm/cc and is given by,

$$D_b = M / V_o$$

Where, D_b = Bulk density (gm/cc) M is the mass of powder (g)
 V_o is the bulk volume of powder (cc)

B. Tapped density (D_t)

Ten grams of powder was introduced into a clean, dry 100ml measuring cylinder. The cylinder was then tapped 100 times from a constant height and tapped volume was read. It is expressed in gm/cc and is given by,

$$D_t = M / V_t$$

Where,
 D_t = Tapped density (gm/cc)
M is the mass of powder (g)
 V_t is the tapped volume of powder (cc)

C. Compressibility index: The compressibility of the powder was determined by the

Carr's compressibility index.

$$CI = \frac{\rho_{\text{tap}} - \rho_{\text{bulk}}}{\rho_{\text{tap}}} \times 100$$

where ρ_{tap} is the tap density and ρ_{bulk} is the bulk density.

Table 4.4: Relation between the Carr's index of powder and its flow characteristics:

Sr.No.	Carr's index	Type of flow
1.	5-15	Excellent
2.	12-15	Good
3.	18-21	Fair
4.	23-30	Poor
5.	33-38	Very poor
6.	>40	Extremely poor

D. Hausner ratio:

Hausner ratio = tapped density / bulk density

Values of Hausner ratio; <1.25: good flow

>1.25: poor flow

If Hausner ratio is between 1.25-1.5, flow can be improved by addition of glidants.

E. Angle of repose (θ)

It is defined as the maximum angle possible between the surface of pile of the powder and the horizontal plane. Fixed funnel method was used. A funnel was fixed with its tip at a given height (h), above a flat horizontal surface on which a graph paper was placed. Powder was carefully poured through a funnel till the apex of the conical pile just touches the tip of funnel. The angle of repose was then calculated using the formula,

$$\theta = \tan^{-1} \left(\frac{h}{r} \right)$$

where,

θ = angle of repose

h = height of pile, r = radius of the base of the pile.

Table 4.5: Comparison between angles of repose and flow property:

Angle of Repose	Flow
<25	Excellent
25 – 30	Good
30 – 40	Moderate (addition of 0.2% glidant required)
>40	Poor

Post Compression Parameters⁵³

A. Thickness and diameter

Control of physical dimension of the tablets such as thickness and diameter is essential for consumer acceptance and tablet uniformity. The thickness and diameter of the tablet was measured using Vernier calipers. It is measured in mm.

B. Hardness

The Monsanto hardness tester was used to determine the tablet hardness. The tablet was held between a fixed and moving jaw. Scale was adjusted to zero; load was gradually increased until the tablet fractured. The value of the load at that point gives a measure of hardness of the tablet. Hardness was expressed in Kg/cm².

C. Friability (F)

Tablet strength was tested by Friabilator USP EF-2. Preweighed tablets were allowed for 100 revolutions (4 min), taken out and were dedusted. The percentage weight loss was calculated by reweighing the tablets. The % friability was then calculated by,

$$F = \frac{(W_{\text{initial}}) - (W_{\text{final}})}{(W_{\text{initial}})} \times 100$$

D. Weight variation test

The weight variation test is carried out in order to ensure uniformity in the weight of tablets in a batch. First the total weight of 20 tablets from each formulation is determined and the average is calculated. The individual weight of the each tablet is also determined to find out the weight variation

E. Uniformity of drug content.

Five tablets of various formulations were weighed individually and powdered. The powder equivalent to average weight of tablets was weighed and drug was extracted in different buffers, the drug content was determined using a UV/Visible Spectrophotometer (Single beam spectrophotometer).

In-vitro release study:

Apparatus	USP XXIV dissolution testing apparatus II (Paddle method)
Dissolution medium	0.1N HCL, 6.8 pH phosphate buffer,
Temperature	37± 0.5 ⁰ C
RPM	50
Vol. withdrawn and replaced	5ml every 1 hour
λ max	279 nm in pH 1.2 buffer and 279 nm in pH 6.8
Blank solution	Buffers used
Duration of study	12 hours
Volume of dissolution media	900ml

Procedure:

The release rate of Ciprofloxacin from tablets was determined using The United States Pharmacopoeia (USP) XXIV dissolution testing apparatus II (paddle type). The dissolution test was performed using 900 ml of pH 1.2, for first 2 hours and followed by phosphate buffer (pH 6.8; 900 mL) for remaining hours at 37.5±0.5⁰C and 50 rpm. A sample (5 ml) of the solution was withdrawn from the dissolution apparatus hourly for 12 hours, and the samples were replaced with fresh dissolution medium. The samples diluted to a suitable concentration with respected dissolution medium. Absorbance of these solutions was measured using a UV-Visible Spectrophotometer (Single beam spectrophotometer). Cumulative percentage of drug release was calculated.

Kinetic Analysis of In-Vitro Release Rates of Controlled Release Tablets

The results of in vitro release profile obtained for all the formulations were plotted in modes of data treatment as follows:-

1. Zero – order kinetic model – Cumulative% drug released versus time.
2. First–order kinetic model – Log cumulative percent drug remaining versus time.
3. Higuchi’s model – Cumulative percent drug released versus square root of time.
4. Korsmeyer equation/Peppas’s model – Log cumulative percent drug released versus log time.

• Zero Order Kinetic

It describes the system in which the drug release rate is independent of its concentration.

$$Q_t = Q_0 + K_0 t$$

Where

Q_t = Amount of drug dissolved in time t

Q_0 = Initial amount of drug in the solution, which is often zero and

K_0 = zero order release constant.

If the zero order drug release kinetic is obeyed, then a plot of Q_t versus t will give a straight line with a slope of K_0 and an intercept at zero.

• First Order Kinetic

It describes the drug release from the systems in which the release rate is concentration dependent.

$$\log Q_t = \log Q_0 + kt / 2.303$$

Where

Q_t = amount of drug released in time t .

Q_0 = initial amount of drug in the solution k = first order release constant

If the first order drug release kinetic is obeyed, then a plot of $\log(Q_0 - Q_t)$ versus t will be a straight line with a slope of $kt / 2.303$ and an intercept at $t=0$ of $\log Q_0$

- **Higuchi Model**

It describes the fraction of drug release from a matrix is proportional to square root of time.

$$M_t / M_\infty = kHt^{1/2}$$

Where

M_t and M_∞ are cumulative amounts of drug release at time t and infinite time,

And

kH = Higuchi dissolution constant reflection formulation characteristics.

If the Higuchi model of drug release (i.e. Fickian diffusion) is obeyed, then a plot of M_t / M_∞ versus $t^{1/2}$ will be a straight line with slope of kH .

- **Korsmeyer-Peppas model (Power Law)**

The power law describes the drug release from the polymeric system in which release deviates from Fickian diffusion, as expressed in following equation.

$$M_t / M_\infty = kt^n$$

$$\log [M_t / M_\infty] = \log k + n \log t$$

Where

M_t and M_∞ are cumulative amounts of drug release at time t and infinite time

(i.e. fraction of drug release at time t),

device, k = constant incorporating structural and geometrical characteristics of CR

n = diffusional release exponent indicative of the mechanism of drug release for drug dissolution.

To characterize the release mechanism, the dissolution data $\{M_t / M_\infty < 0.6\}$ are evaluated.

A plot of $\log \{M_t / M_\infty\}$ versus $\log t$ will be linear with slope of n and intercept gives the value of $\log k$.

Antilog of $\log k$ gives the value of k .

Peppas used the n value in order to characterize different release mechanisms as shown in the table below

Table 4.8: Mechanism of Drug Release as per Korsmeyer Equation/Peppas's**Model:**

S. No	Release exponent	Drug transport mechanism	Rate as a function of time
1	0.5	Fickian diffusion	$t^{-0.5}$
2	$0.45 < n = 0.89$	Non-Fickian transport	t^{-n-1}
3	0.89	Case II transport	Zero order release
4	Higher than 0.89	Super case II transport	t^{-n-1}

RESULTS AND DISCUSSIONS

PREFORMULATION STUDIES

Determination of melting point

The melting point of Ciprofloxacin was found to be 312°C which compiled with BP standards, indicating purity of the drug sample, which was determined by capillary method.

Solubility studies:

Table: Solubility studies of Ciprofloxacin

Solvent	% Solubility
1.2 pH buffer (0.1N HCL)	0.245±0.045
7.4 pH Phosphate buffer	0.586±0.071
6.8 pH Phosphate buffer	0.683±0.065

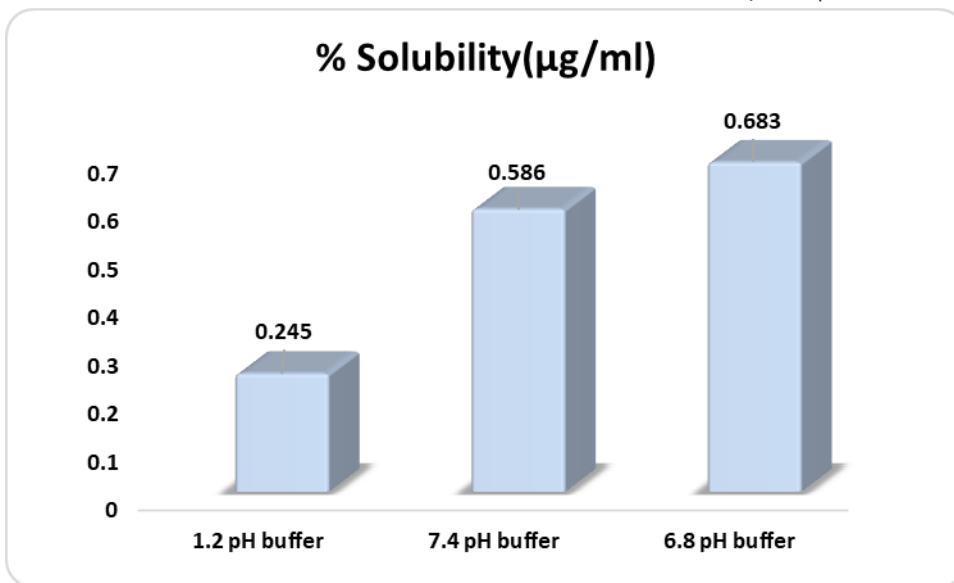


Fig5.1: Solubility studies of Ciprofloxacin

Discussion: From the above solubility studies, it was observed that among that 3 buffer solutions (0.1N HCL i.e pH 1.2, 7.4 pH Phosphate buffer and 6.8 pH phosphate buffer) the drug was soluble freely in 6.8 pH buffer.

FTIR studies:

Drug-Excipient compatibility studies:

The IR spectrum of pure drug was found to be similar to the standard spectrum of Ciprofloxacin.

From the spectra of Ciprofloxacin, combination of Ciprofloxacin with polymers, it was observed that all characteristic peaks of Ciprofloxacin were not altered and present without alteration in the combination spectrum, thus indicating compatibility of the drug and polymers. FTIR spectra of Ciprofloxacin, and Optimized formulation are shown in Figure respectively.

FTIR Spectra of Pure drug:

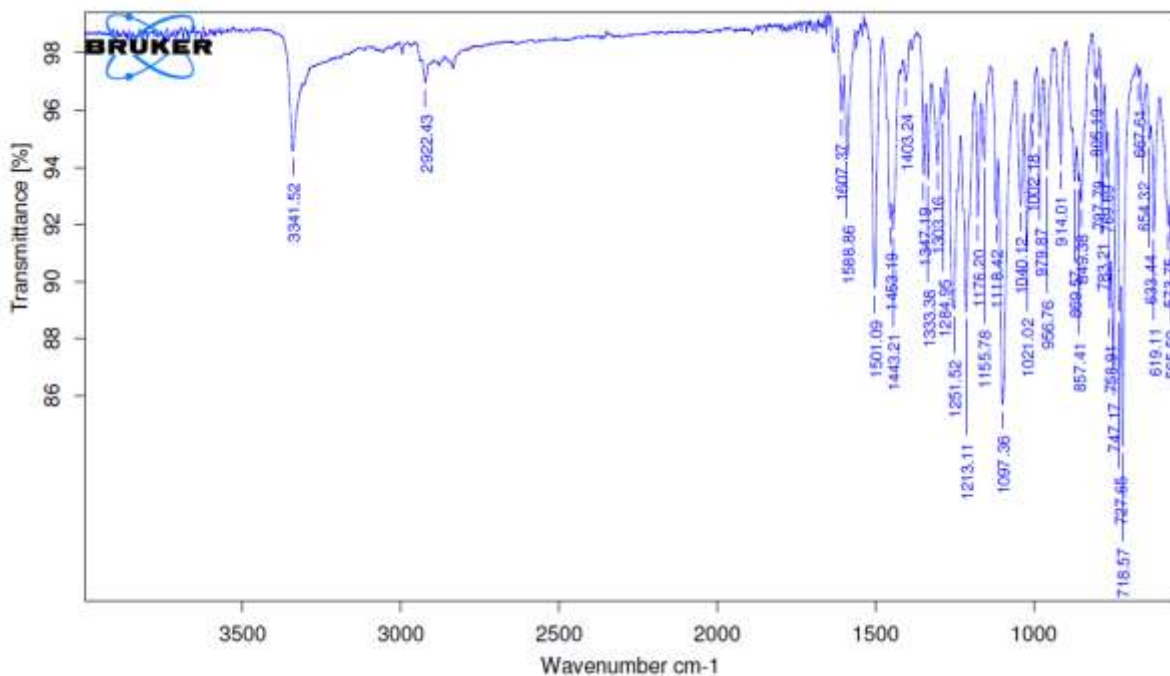


Fig.5.7 FTIR Spectra of Pure drug

FTIR Spectra of Drug and Excipients:

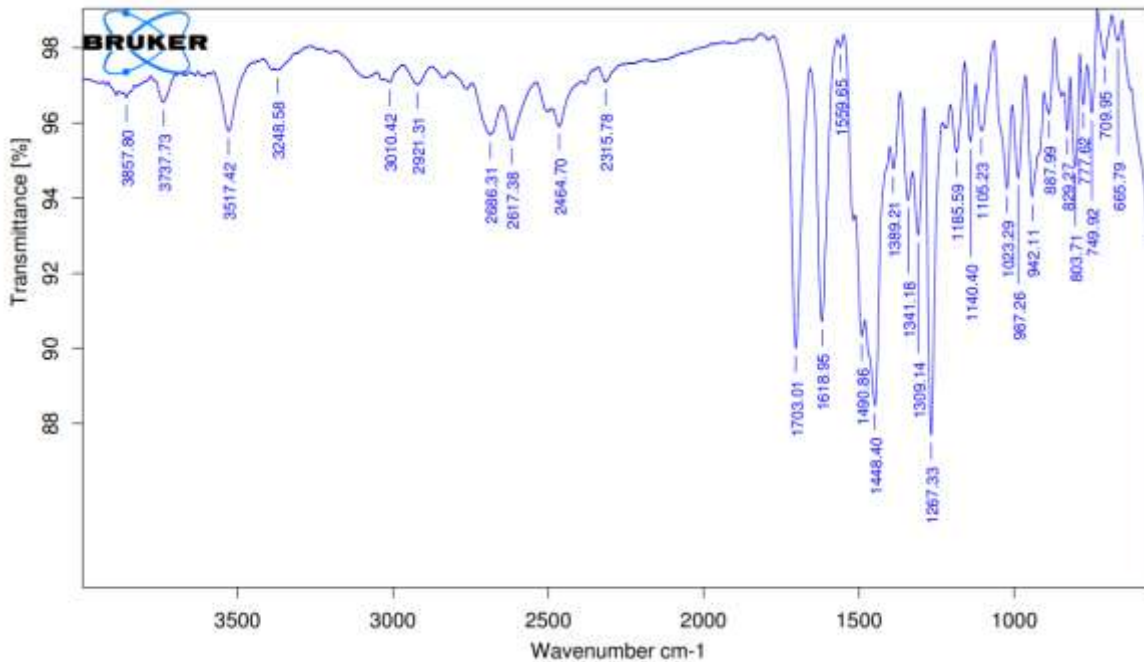


Fig.5.7.1FTIR Spectra of Drug and Excipients

Discussion: From the drug excipients compatibility studies we observe that there are no interactions between the pure drug and (drug+ excipients) which indicates there are no physical changes.

UV Determination:

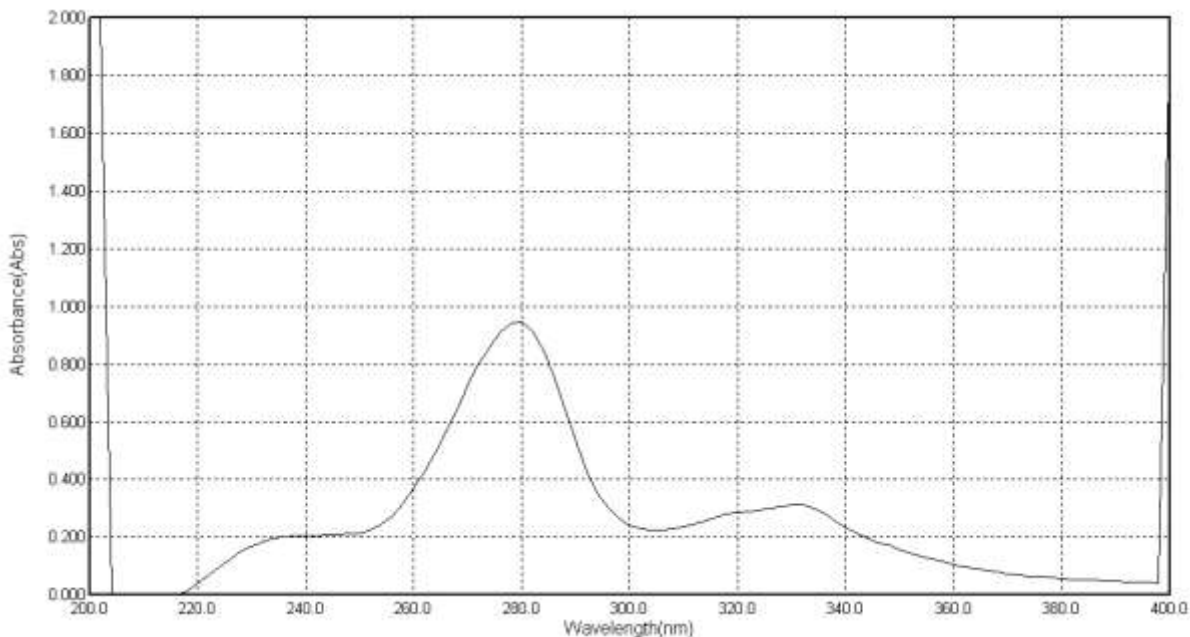
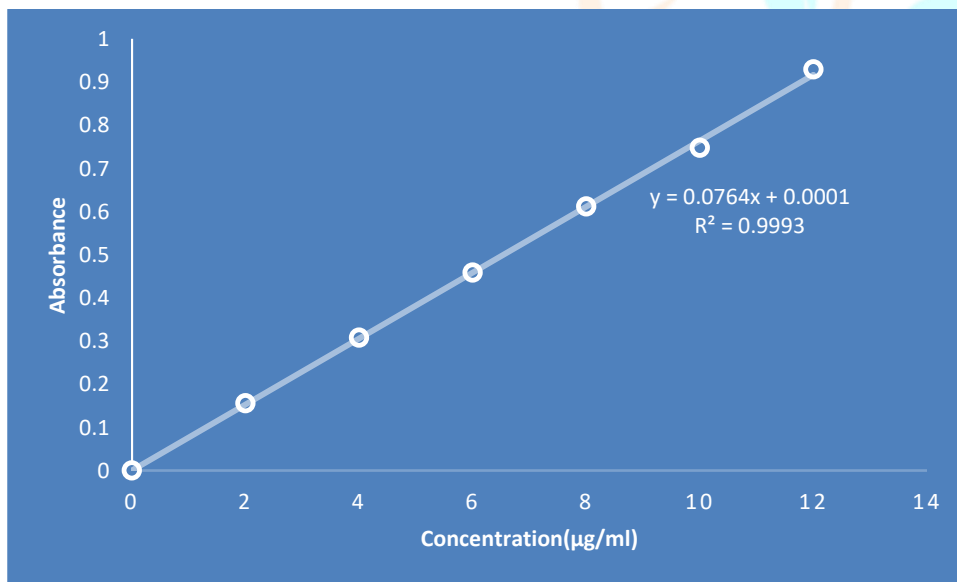


Fig.5.6.Uv spectrum of Ciprofloxacin

Discussion: A solution of Ciprofloxacin containing the concentration of 8 µg/ml was prepared in 6.8 pH buffer and UV spectrum was taken using Single Beam Spectrophotometer (YIS-294). The solution was scanned in the range of 200–400 nm. The maximum absorbance was found to be at 279 nm.

Standard Calibration Curve of Ciprofloxacin in pH 1.2 Buffer:**Table: Standard Calibration Curve of Ciprofloxacin in pH 1.2 Buffer**

Concentration($\mu\text{g/ml}$)	Absorbance
0	0
2	0.176
4	0.297
6	0.458
8	0.624
10	0.792
12	0.932

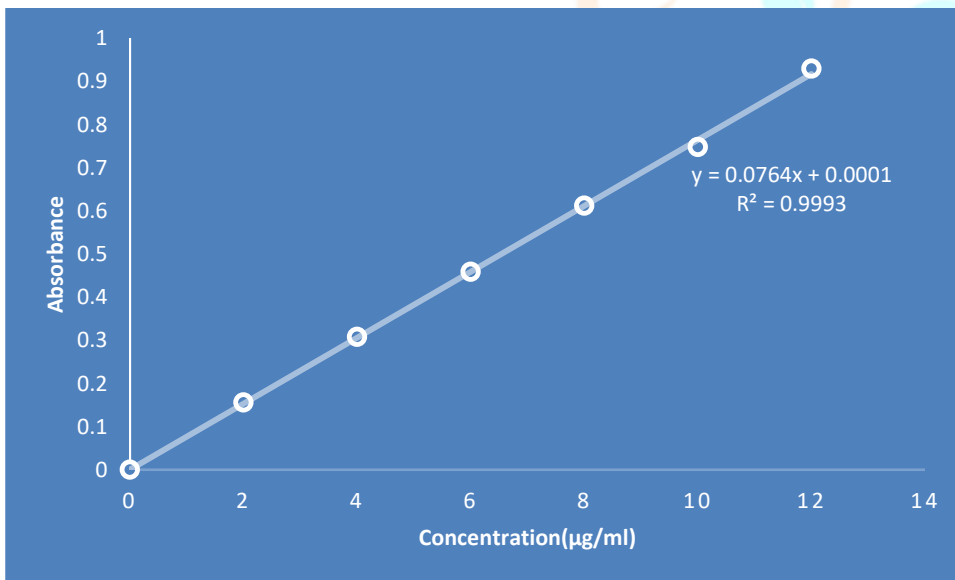
**Fig : Standard calibration curve of Ciprofloxacin in pH 1.2****Discussion:**

The standard calibration curve shown obeys Beer's and Lambert's law in the concentration range of 0 to 12 $\mu\text{g/mL}$. A standard graph was plotted by keeping the known concentration on X-axis and obtained absorbance on Y-axis.

5.3.3. Standard Calibration Curve of Ciprofloxacin in pH 6.8:

Table:7.4: Standard Calibration Curve of Ciprofloxacin in pH 6.8

Concentration($\mu\text{g/ml}$)	Absorbance
0	0
2	0.155
4	0.307
6	0.458
8	0.612
10	0.747
12	0.929

**Fig : standard calibration curve Ciprofloxacin in pH 6.8 Phosphate Buffer****Discussion:**

The standard calibration curve shown R^2 value 0.999 which is near to 1 shows the linearity, through that the drug obeys Beer's and Lambert's law in the concentration range of 0 to 12 $\mu\text{g/mL}$. A standard graph was plotted by keeping the known concentration on X-axis and obtained absorbance on Y-axis.

Evaluation of Ciprofloxacin controlled release matrix Tablets**Table:5.5. Pre Compression Parameters of Ciprofloxacin controlled release Tablets:**

FC	Bulk density	Tapped density	Carr's index	Hausner's ratio	Angle of Repose
F1	0.485 \pm 0.004	0.535 \pm 0.006	18.45 \pm 1.25	1.21 \pm 0.12	25.25 \pm 0.54
F2	0.454 \pm 0.006	0.526 \pm 0.004	17.29 \pm 1.29	1.24 \pm 0.16	28.48 \pm 0.26

F3	0.467±0.004	0.529±0.001	16.46±1.37	1.28±0.28	27.26±0.36
F4	0.476±0.006	0.534±0.002	15.12±1.23	1.21±0.34	29.16±0.18
F5	0.468±0.002	0.529±0.008	17.26±1.19	1.16±0.16	26.48±0.24
F6	0.474±0.004	0.542±0.004	16.48±1.15	1.18±0.28	27.02±0.23
F7	0.485±0.002	0.581±0.006	14.26±1.28	1.17±0.19	28.39±0.29
F8	0.497±0.007	0.598±0.004	12.77±1.24	1.11±0.25	24.12±0.24
F9	0.456±0.004	0.587±0.002	19.26±1.16	1.19±0.24	28.48±0.56
F10	0.474±0.001	0.585±0.007	18.29±1.12	1.18±0.18	27.58±0.18
F11	0.479±0.001	0.557±0.002	17.47±1.24	1.15±0.47	26.74±0.45
F12	0.487±0.005	0.575±0.001	15.64±1.18	1.14±0.51	27.03±0.89

Discussions:

- The angle of repose of different formulations (F1-F12) was found to be in the range of 24.12±0.24 to 29.16±0.18. But the formulation F9 having the excellent flow property with 24.12±0.24. So, it was confirmed that the flow property of blends was free flowing.
- The bulk density of blend was found between 0.454±0.006 to 0.497±0.007.
- The Tapped density was found between 0.529±0.001 to 0.598±0.004. These values indicate that the blends had good flow property.
- Carr's index for all the formulations was found to be between 18.45±1.25 to 12.77±1.24 and Hausner's ratio from 1.28±0.28 to 1.11±0.25 which reveals that the blends have good flow character.

5.7 PostCompressionParameters ofCiprofloxacin controlled release tablets:

Table 5.6: Physical properties oftabletfomulation(F1 toF12):

FC	Thickness (mm)	Hardness (kg/cm²)	Friability (%)	Weight Variation (mg)	Drug Content (%)
F1	5.16±1.66	7.58±0.74	0.72±0.18	602.29±1.54	96.43±0.42
F2	5.72±0.78	7.47±0.62	0.66±0.82	601.02±0.26	95.25±0.26
F3	5.32±0.88	8.45±0.95	0.54±0.66	599.56±0.54	93.63±0.39
F4	5.58±0.74	8.67±0.76	0.56±0.38	598.28±0.11	95.18±0.25
F5	5.37±0.82	8.28±0.88	0.69±0.65	600.64±0.78	93.25±0.26
F6	5.69±0.79	8.56±0.94	0.56±0.26	601.51±0.86	95.43±0.38
F7	5.57±0.72	8.62±0.37	0.48±0.72	598.24±0.24	97.67±0.45
F8	5.68±0.36	8.98±0.72	0.45±0.58	600.47±0.54	98.35±0.26

F9	5.85±0.84	8.89±0.49	0.94±0.45	601.12±0.18	94.74±0.38
F10	5.87±0.84	7.48±0.84	0.87±0.84	602.14±0.84	96.47±0.84
F11	5.12±0.84	8.25±0.84	0.68±0.84	598.15±0.84	95.47±0.84
F12	5.29±0.84	8.79±0.84	0.59±0.84	599.52±0.84	97.68±0.84

Discussion:

- Thickness of the Ciprofloxacin tablets were found to be in the range of 5.12±0.84 mm to 5.85±0.84 mm.
- Hardness of the Ciprofloxacin tablets were found to be in the range of 7.47±0.62 to 8.98±0.72 kg/cm².
- Friability of the Ciprofloxacin tablets were found to be in the range of 0.45±0.58 to 0.94±0.45 %.
- The Weight Variation of the Ciprofloxacin tablets were found to be in the range of 598.15±0.84 to 602.29±1.54 mg.
- Drug content of the Ciprofloxacin tablets were found to be in the range of 93.63±0.39% to 98.35±0.26%.

5.8 In-vitro drug releasestudies:

In-vitro drug releasestudies were carried out using USPXXII dissolution apparatus type II (Lab India DS 8000) at 50 rpm. The dissolution medium consisted of 900 ml of buffer, maintained at 37±0.5 °C. The drug release at different time intervals was measured using an ultraviolet visible spectrophotometer (PG Instruments). The study was performed in triplicate

Table.5.7: In vitro dissolution studies with formulation F1 to F4

Time (hrs)	F1	F2	F3	F4
0	0	0	0	0
1	25.47 ±1.48	29.47 ±1.45	17.45 ±1.36	12.47 ±1.24
2	34.47 ±1.19	43.77 ±1.22	25.46 ±1.36	26.45 ±1.21
3	48.47 ±1.45	55.42 ±1.16	35.14 ±1.36	34.19 ±1.25
4	59.87 ±1.47	67.41 ±1.01	49.87 ±1.46	45.48 ±1.17
5	74.58 ±1.65	75.92 ±1.24	57.82 ±1.22	57.15 ±1.28
6	88.47 ±1.25	83.72 ±1.36	64.46 ±1.14	66.41 ±1.15
7	96.48 ±1.62	89.35 ±1.87	72.84 ±1.22	72.16 ±1.42
8		92.08 ±1.11	79.48 ±1.54	81.45 ±1.21
9		98.47 ±1.24	87.25 ±1.28	88.68 ±1.10
10			98.36 ±1.26	92.14 ±1.48
11				98.58 ±1.44
12				

Table.5.7: In vitro dissolution studies with formulation F5 to F8

Time (hrs)	F5	F6	F7	F8
0	0	0	0	0
1	31.82 ±1.13	24.48 ±1.14	16.72 ±1.21	14.57 ±1.48
2	47.26 ±1.22	33.78 ±1.26	28.42 ±1.67	28.24 ±1.47
3	59.24 ±1.38	42.14 ±1.42	31.46 ±1.22	37.48 ±1.54
4	66.84 ±1.42	54.81 ±1.19	39.58 ±1.44	45.24 ±1.29
5	75.86 ±1.19	66.46 ±1.04	48.82 ±1.51	56.34 ±1.61
6	83.92 ±1.58	75.98 ±1.54	57.77 ±1.13	63.48 ±1.24
7	91.64 ±1.22	81.89 ±1.12	68.21 ±1.27	70.19 ±1.37
8	98.86 ±1.04	89.21 ±1.36	76.48 ±1.25	79.18 ±1.45
9		93.15 ±1.28	82.46 ±1.57	85.29 ±1.36
10		98.26 ±1.36	91.49 ±1.64	91.42 ±1.28
11			98.16 ±1.95	95.43 ±1.28
12				99.41 ±1.25

Table.5.7: In vitro dissolution studies with formulation F9 to F12

Time (hrs)	F9	F10	F11	F12
0	0	0	0	0
1	29.82 ±1.48	22.86 ±1.14	17.48 ±1.14	13.58 ±0.14
2	35.26 ±1.27	31.78 ±1.26	26.78 ±1.26	26.47 ±1.24
3	48.24 ±1.34	39.14 ±1.42	34.14 ±1.42	34.45 ±1.21
4	55.84 ±1.54	47.81 ±1.19	58.81 ±1.19	48.19 ±1.25
5	68.88 ±1.42	56.46 ±1.04	62.46 ±1.04	53.48 ±1.17
6	75.95 ±1.14	65.98 ±1.54	68.98 ±1.54	64.15 ±1.28
7	87.47 ±1.25	78.89 ±1.12	75.89 ±1.12	72.41 ±1.15
8	98.24 ±1.15	84.21 ±1.36	80.21 ±1.36	78.16 ±1.42
9		91.15 ±1.45	88.15 ±1.28	83.45 ±1.21
10		97.26 ±1.26	93.26 ±1.36	87.68 ±1.10
11			98.74 ±1.36	93.14 ±1.48
12				97.58 ±1.44

In Vitro Drug Release Studies:

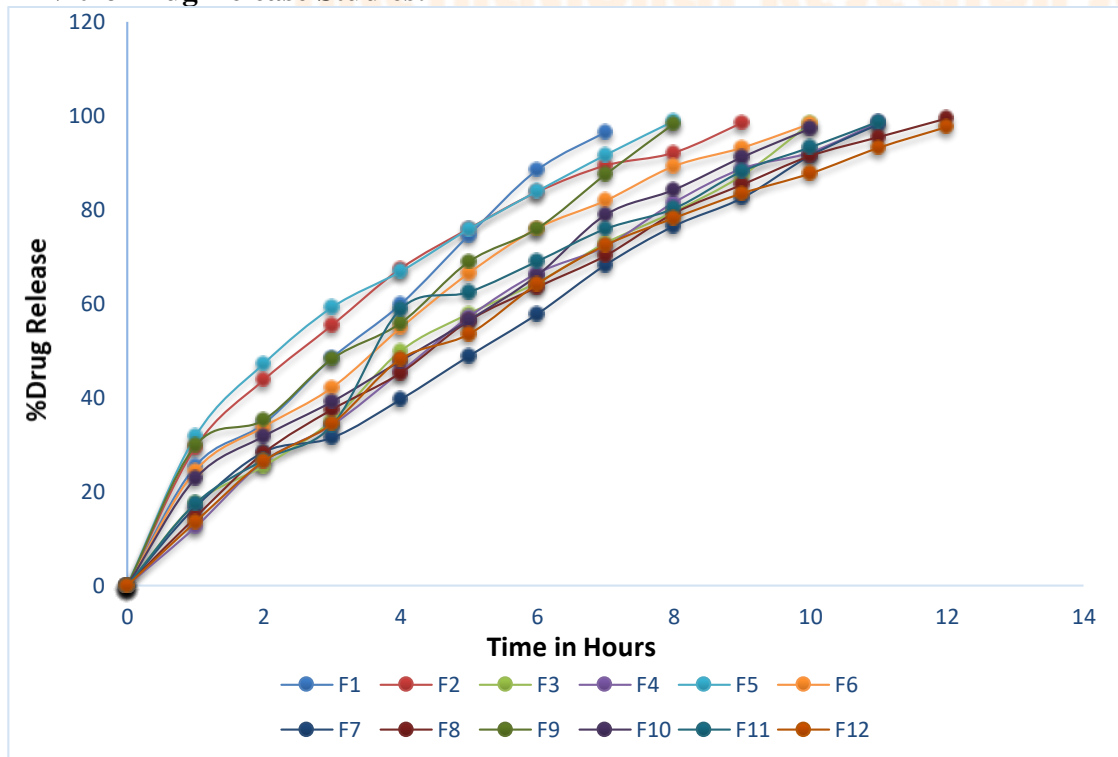


Fig5.8:In Vitro Drug Release Studies Of F1-F12 Formulations

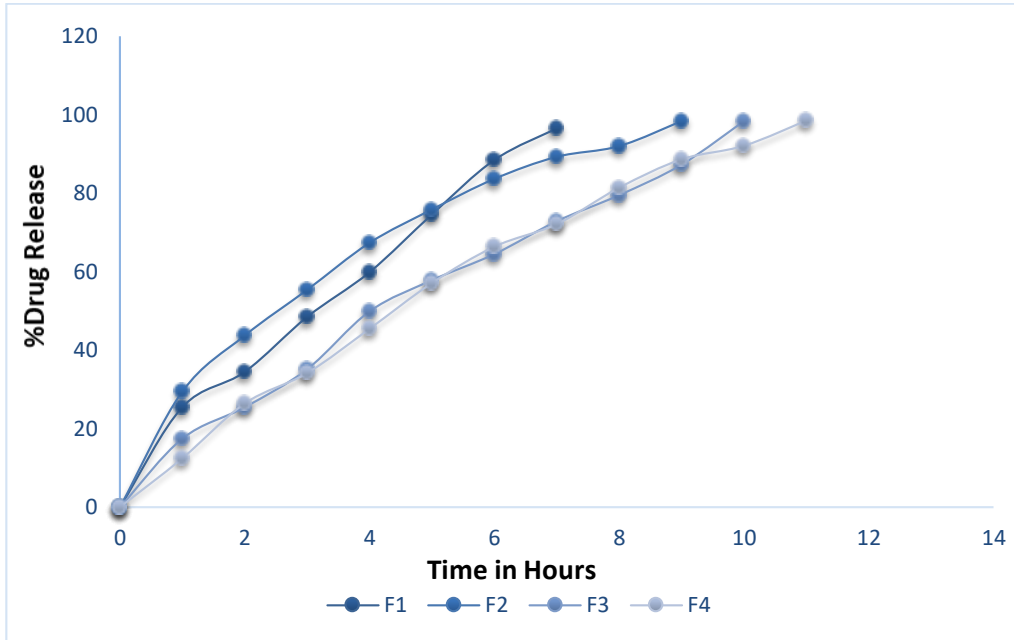
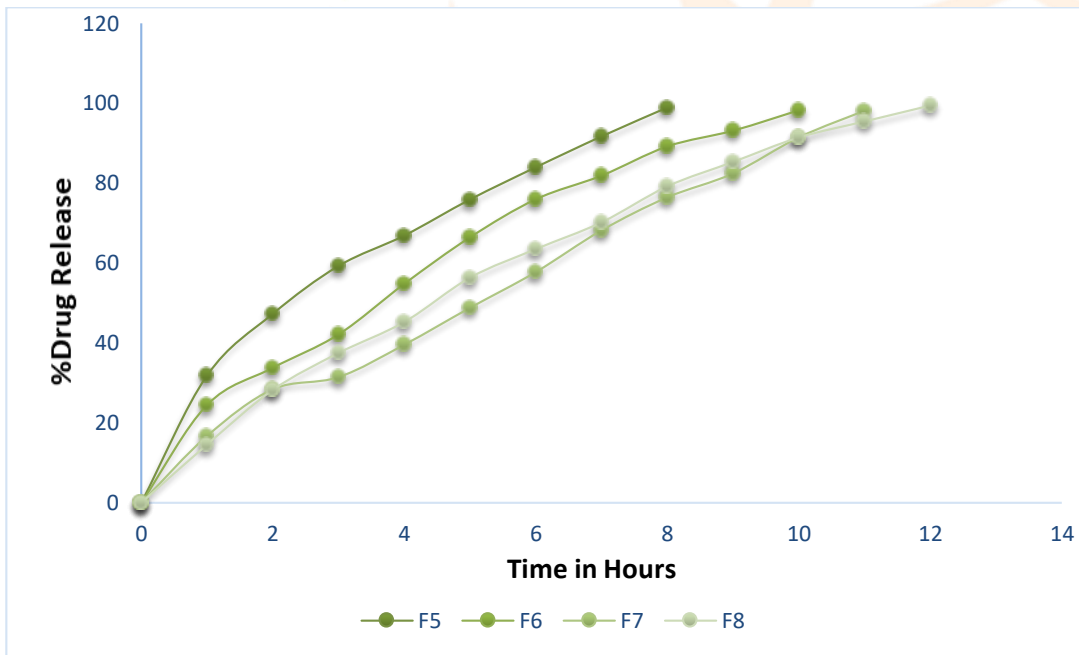
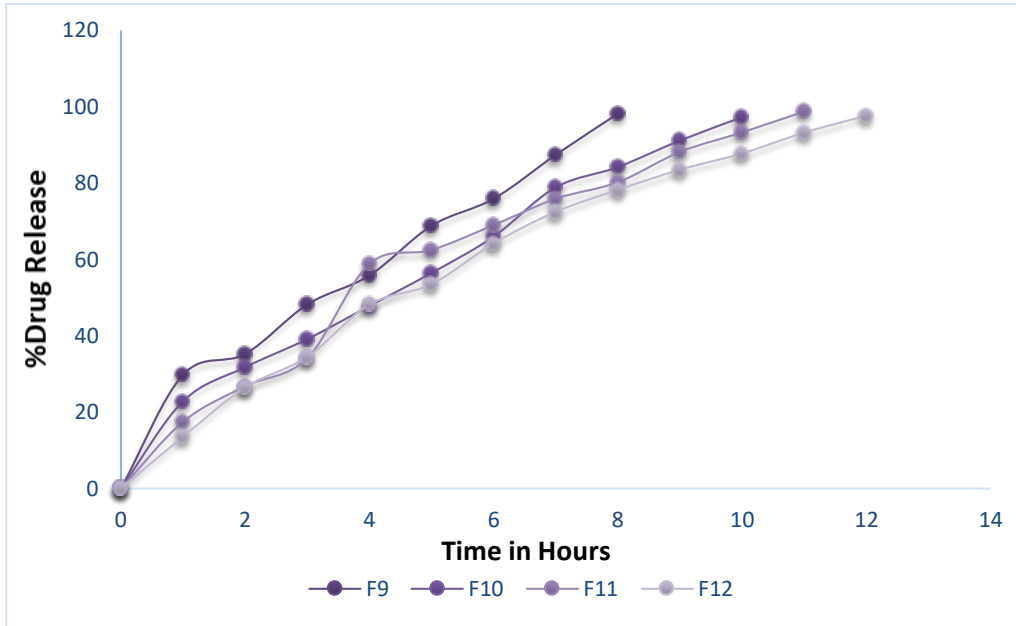


Fig:5.9:In Vitro Drug Release Studies Of F1-F4 Formulations



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Fig:5.10:In Vitro Drug Release Studies Of F5-F8 Formulations**Fig:5.10:In Vitro Drug Release Studies Of F9-F12 Formulations****Discussion:**

From the in vitro drug release studies of Ciprofloxacin controlled release tablets using Xanthan Gum, Carbopol 940 and Ethyl cellulose in three different polymer ratios using lactose as a diluent, MCC as a binder and PVP K30 as filler. From Formulations, the F1-F4 were formulated using xanthan gum in four different ratios like 100mg, 150mg, 200mg and 250mg the drug release of F1 was 96.48 ± 1.62 at the end of 7th hour, F2 was 98.47 ± 1.24 at the end of 9th hour and F3 was 98.36 ± 1.26 at the end of 10th hour and F4 releases $98.58 \pm 1.44\%$ at the end of 11th hour.

So, when the polymer concentration increases the drug release time was increased but F4 showing the highest release at the end of 11th hour only. So further dissolution was takes on the polymer Carbopol 940.

The formulation F5-F8 were formulated by using Carbopol 940 in four different ratios like 100mg, 150mg, 200mg and 250mg, the drug release of F5 was 98.86 ± 1.04 at the end of 8th hour, F6 was $98.26 \pm 1.36\%$ at the end of 10th hour and F7 was 98.16 ± 1.95 at the end of 11th hour and F8 releases $99.41 \pm 1.25\%$ at the end of 12th hour

While the formulation done with the Carbopol 940 shows the higher amount of drug released at the end of 12th hour. So further studies were done with Ethyl cellulose.

The formulations from F9-F12 were formulated by using Ethyl cellulose in four different ratios like 100mg, 150mg, 200mg and 250mg, so the drug release of F9 was 98.24 ± 1.15 at the end of 8th hour, F10 was $97.26 \pm 1.26\%$ at the end of 10th hour and F11 was 98.74 ± 1.36 at the end of 12th hour and F12 releases $97.58 \pm 1.44\%$ at the end of 12th hour

In this turn, with the polymer Carbopol 940 the formulation F8 shows the better results up to 12th hour. Hence the formulation F8 was identified as the Optimized formulation. So further Release kinetics was done to this formulation.

5.9. Drug Release Kinetics:

Zero Order:

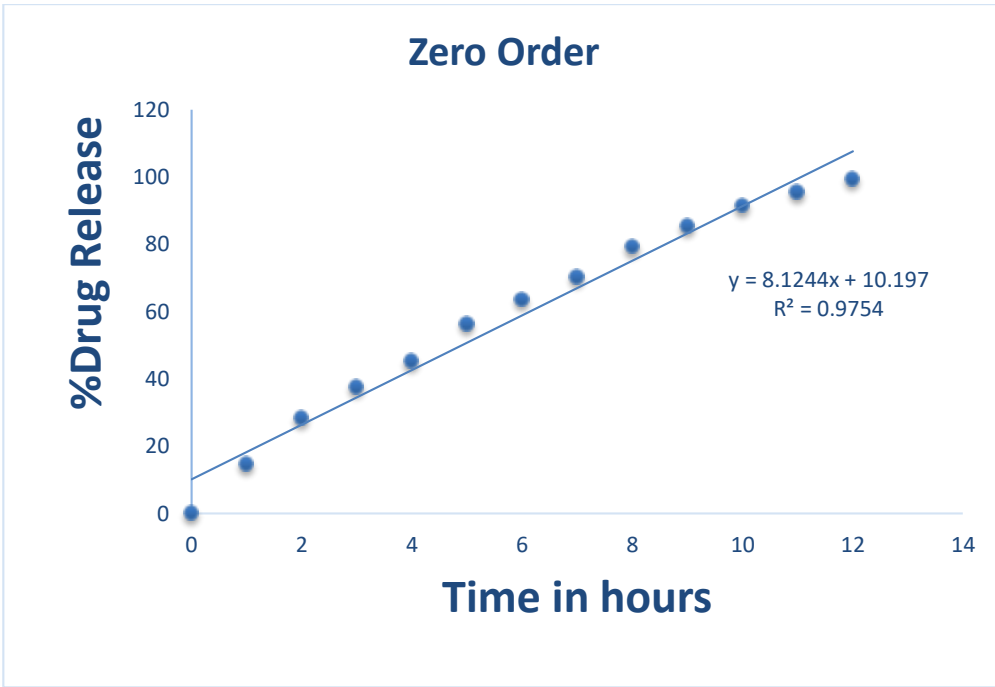


Fig: 5.12. Zero order graph of optimized formulation(F8)

First Order:

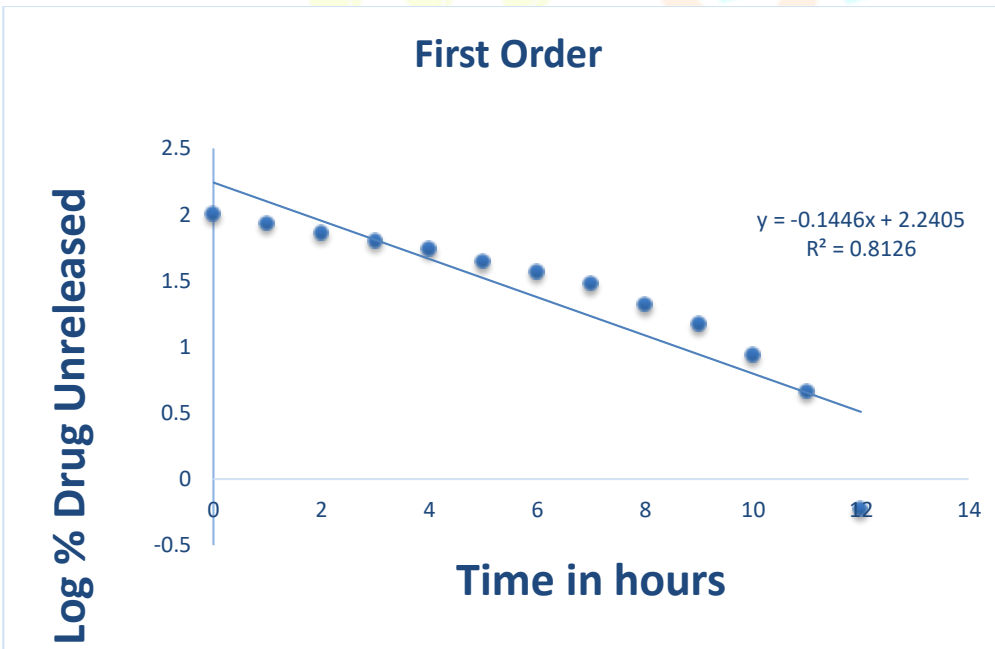


Fig:5.13. First order graph of optimized formulation(F8)



Higuchi Plot:

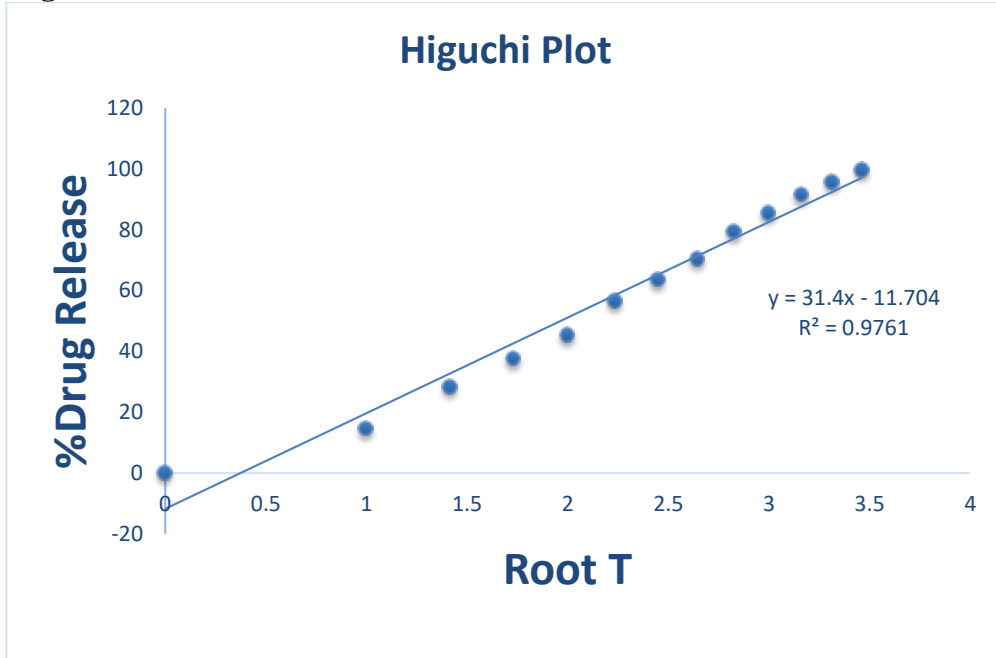


Fig:5.14.Higuchi graph of optimized formulation(F8)

Peppas Plot:

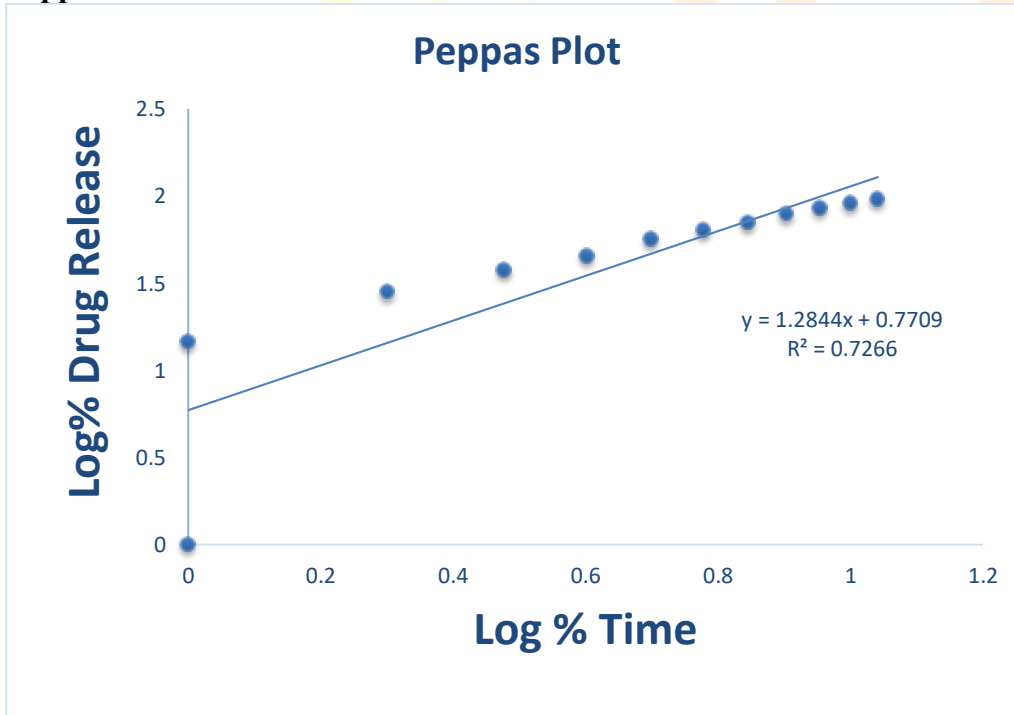


Fig:5.15.Peppas graph of optimized formulation(F8)

Table:5.8.Drug release kinetics

R ² values					n values
Formulation	Zero order	First order	Higuchi	Korsmeyer - Peppas	Korsmeyer-Peppas (n)
F8	0.975	0.812	0.972	0.726	1.284

Discussions:

The invitro dissolution data for the optimized formulation F9 were fitted in different kinetic models i.e, zero order, first order, Higuchi and korsmeyer-peppas equation. Optimized formulation F9 shows R^2 value 0.975. As its value nearer to the '1' it is conformed as it follows the Zero order release. The mechanism of drug release is further confirmed by the korsmeyer and peppas plot, if $n = 0.45$ it is called Case I or Fickian diffusion, $0.45 < n < 0.89$ is for anomalous behavior or non-Fickian transport, $n = 0.89$ for case II transport and $n > 0.89$ for Super case II transport. The 'n' value is 1.284 for the optimized formulation (F8) i.e., n value was > 0.89 this indicates super case transport. The release kinetics for the optimized formula are shown in table.

SUMMARY

In this study, controlled release tablets of Ciprofloxacin were prepared by Direct compression method, using Xanthan Gum, Carbopol 940 and Ethyl cellulose polymers. The pre compression and post compression parameters show that the values were found to be acceptable within the range. FT-IR studies revealed that the drug and excipients used weren't have any interactions. The drug-polymer ratio was found to influence the release of drug from the formulations.

Different parameters like hardness, friability, weight variation, drug content uniformity, *in-vitro* drug release were evaluated in all the parameters the optimized formulation F9 yields best results and among in all 12 formulations F8 formulation containing 250mg of Carbapol 940 controls the drug release up to 12 hours. So Carbapol 940 was considered to be suitable for the formulation of Ciprofloxacin controlled release tablets. Based on these results formulation F8 was found to be the most promising formulations. The regression coefficient (R^2) of Zero order of Optimized formulation F8 shows R^2 value 0.975. The 'n' value is 1.284 for the optimized formulation (F8) i.e., n value was > 0.89 this indicates super case transport

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